Synthesis and QSAR Studies of Novel Pyrazoline Derivatives as Antiproliferative Agent

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ABSTRACT

Background: Regardless of recent advances in the biological processes leading to the development of cancer, there is still a need for new and effective agents to help bring this disease under control. The revelation of the anticancer properties of pyrazoline makes this scaffold remarkable for research and development as an anticancer. **Methods:** A series of synthetic chalcones and pyrazoline derivatives were synthesized, characterized IR, ¹H NMR and mass spectral analysis and evaluated for their Anticancer activity on MCF-7 cell line through SRB assay. A QSAR study was performed on thirty-five compounds of known series having pyrazoline nucleus using Accelrys Discovery Studio Vs 2.1.0.8130. **Results:** We found that significant inhibitory activity among all the synthesized compound **1B** showing better anticancer activity having IC₅₀ value 0.010 μ M as compared to standard drug letrozole. So predicted molecules from the QSAR study show significant activity. **Conclusion:** We would conclude that compound **1B** might serve as a potential antiproliferative agent.

Key words: Pyrazoline, QSAR, SRB assay, MCF-7, Antiproliferative activity.

INTRODUCTION

Cancer is known medically as a malignant neoplasm which is a broad group of various diseases involving unregulated cell growth. In cancer cells are dividing and grow uncontrollably resulted in malignant tumours and then invade nearby parts of the body.1 Behind cardiovascular disease, cancer is the second leading cause of death.^{2,3} Receptor protein tyrosine kinases transduce signal pathways that regulate cell division and differentiation. epidermal growth factor receptor (EGFR) kinase plays a key role in growth factors that have been identified as being important in cancer^{4,5} through studies of breast cancer suggest that 25-30% of cases showed elevated expression of EGFR. Overexpression also was seen in lung cancer, ovarian cancer and prostate cancer.^{6,7} Compounds that inhibit EGFR kinase activity after binding cognate ligands are of potential interest as new therapeutic anticancer agents.8 Consequently, inhibition of the proliferative pathway is considered

an effective approach in the battle against cancer. Different structural modifications of the pyrazoline motif have been made to explore its features/ properties and biological potential.9 Pyrazoline and its derivatives are recognized to possess a wide array of biological activities. The pyrazoline scaffold is the fundamental structure of many bioactive compounds. Pyrazoline shows different activities like anti-inflammatory activity,¹⁰ antiviral activity,¹¹ antifungal activity,¹² anticancer¹³ and antimicrobial activity.14 Significant attention was given to pyrazoline as an anticancer drug after the discovery and development of new anticancer agents. Quantitative structure-activity relationships (QSAR) and quantitative structure properties relationships (QSPR) play a central role in this effort and those methods are unquestionable of great importance in modern chemistry and biochemistry.¹⁵ The concept of QSAR/QSPR is to transform

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searches for compounds with desired properties using chemical intuition and experience into a mathematically quantified and computerized form. QSAR attempts to correlate structural, chemical, statistical and physical properties with biological activity by various approaches. QSAR models are scientific credible tools for predicting and classifying biological activities of untested chemicals. QSAR is an essential tool for lead development (optimization); a growing trend is to use QSAR early in the drug discovery process as a screening and enrichment tool to eliminate from further development those chemicals lacking "drug-like" properties or those chemicals predicted to elicit a toxic response.¹⁶

MATERIALS AND METHODS Materials and Instrumentation

Melting points were determined using a VEEGO make microprocessor-based melting point apparatus having a silicone oil bath and are uncorrected. IR spectra (wavenumbers in cm⁻¹) were recorded on a BRUKER ALPHA FT-IR spectrophotometer using Potassium bromide discs. NMR spectra were recorded on BRUKER AVANCE II 400 MHz instrument in DMSO with TMS as an internal standard for ¹H NMR. Chemical shift values are mentioned in δ , ppm. Mass spectra were recorded on Shimadzu LC-MS 2010 spectrometer. Chromatographic separations were performed on columns using silica gel 100-200 mesh. The progress of all reactions was monitored by TLC on 2 cm X 5 cm pre-coated silica gel 60 F₂₅₄ (Merck) plates of a thickness of 0.25 mm. The chromatograms were visualized under UV 254 nm and/or exposure to iodine vapors. All reagents used were of analytical reagent grade, obtained from LOBA chemicals, SDFCL, Spectrochem and Sigma Aldrich. Chemicals and solvents were purified by general laboratory techniques before use. All moisturefree operations were performed in oven-dried glassware. The anti-cancer activity of all synthesized compounds will be evaluated for *in-vitro* measurement by MCF-7 cell line through SRB assay. QSAR study of pyrazoline as anticancer moiety was performed using Accelrys discovery studio client (Version 2.1) as the modelling tool. A total of 35 selected molecules were considered for the development of the QSAR model. The statistical analysis performed revealed following observations; $r^2 = 0.884$, $r^2 cv$ (Cross validated r^2) = 0.883, $r^2 pred$ = 0.99 and $r^2m = 0.61$. A total of 13 descriptors pruned on the study explained the structural correlation of pyrazoline. The model developed can be used to predict the bioefficacy of substituted pyrazoline derivatives as

an anticancer agent. Based on the QSAR study predicted molecules with different substituents are synthesized as per the following scheme Figure 1.

Experimental

A data set of pyrazoline derivatives reported by Ping *et al.* studied the synthesis and anticancer activity of 1-(5-(substitutedphenyl)-3-(substitutedphenyl)-4,5-dihydropyrazol-1yl) ethanone derivatives were taken in form of pIC_{50} . The total set of 35 compounds was randomly divided into a training set and a test set of 25 and 10 compounds, respectively.¹⁷ Chemical structures of the compounds along with the biological activity are shown in Table 1 and Table 2.

In this study, the 2D structure of the molecules was sketched on Chemdraw ultra 4.0 and the QSAR study of these series was done by Accelrys Discovery Studio Vs 2.1.0.8130. Discovery Studio is a single, powerful, easyto-use, graphical interface for drug design and protein modeling research. Discovery Studio 2.1 combines established gold-standard applications such as Catalyst, Modeler and CHARMm that have years of proven results and utilizes cutting-edge science to address the drug discovery challenges of today. Discovery Studio 2.1 is built on the Sci Tegic Enterprise Server platform to seamlessly integrate protein modeling, pharmacophore analysis, virtual screening and third-party applications.

All the structures first minimized by using energy minimization after that calculate molecular property of all minimized structure. Now based on all molecular property calculate the correlation matrix which gives a good correlation between two descriptors. A common practice in building QSAR models is to select descriptors that are not inter-correlated and show a lesser degree of multi-collinearity. This is achieved by determining the correlation between the independent variable (descriptor) of interest and other variables in the regression equation, with value 1 implying the highest multi-colinearity. Variance inflation factor (VIF) analysis is also determined and it should be less than 10. Selections of molecular descriptors for the QSAR study were carried out using the correlation matrix in excel and variance inflation factor analysis. The total numbers of 125 descriptors were calculated which included electronic, spatial, structural, thermodynamic, topological and geometric descriptors but some of them were rejected as they were highly correlated. Remaining descriptors were used to derive a QSAR model using the GFA technique. All the descriptors were calculated using PC based software, Discovery Studio 2.1 (Accelrys Inc, USA). Total 13 molecular descriptors such as Initial RMS Gradient, Improper Energy, Molecular_

FractionalPolarSASA, SIC, Dipole_Y, Jurs_RPCS, Minimized_Energy, Strain_Energy, RadofGyration, PMI_Z, Shadow_XY, Shadow_YZ, Shadow_Zlength. Table 3. One molecular property correlates with another molecular property which indicates such molecular property has a significant effect on molecules.QSAR models were generated using the GFA technique in Discovery studio 2.1 (Accelrys Inc, USA) shown in Table 4. The statistical qualities of models generated were judged by parameters such as Friedman's lack of fit measure (LOF), the square of regression coefficient r^2 , adjusted r²adj, cross-validated r²cv and standard deviation(s) as shown in Table 5.

There are two techniques to determine the reliability of the generated models, internal and external validation. Internal validation uses the dataset from which the model is derived and is required to check internal consistency and stability. To determine the quality of the model internally, cross-validation (cv) techniques are extensively employed. Cross-Validation methods employed for internal validation are Leave-One-Out, Leave-Some-Out, or Leave- Many-Out. The quality of the model is analyzed by the value of the correlation coefficient of the cross-validation procedure, that is, r²cv. The commonly accepted value for a satisfactory QSAR model is $r^2cv > 0.5$. The most common approach of validation is to examine the residuals, i.e., calculated from the difference between observed and predicted biological activity. Lower the residual values better than the model. (Table 6, Figure 2)

The residue of the training set indicates the linearity of the model. The value of the residue gives an idea about the outlier of the selected series of the pyrazoline derivatives (Figure 2). More value of the residue deviated the biological activity.

External Validation

Any model with excellent statistical characteristics (like r^2 , r^2cv , F-value) and satisfactory internal predictions may lack the true relationship between molecular descriptors and target property. To avoid chance correlation, a reliable validation procedure must be carried out. The ultimate validation of the model is examined utilizing external validation. The quality of a QSAR model is

Table	Table 1: Structures and biological activities of training set compounds.								
Sr. No.	Comp No.	<i>R</i> ₁	R ₂	IC ₅₀	pIC₅₀				
1	1a	3,4-di methyl	4-F	0.83	6.080922				
2	1b	3,4-di methyl	4-Cl	1.36	5.866461				
3	1c	3,4-di methyl	4-Br	2.17	5.66354				
4	1e	3,4-di methyl	4-OCH ₃	0.07	7.154902				
5	1g	3,4-di methyl	4-NO ₂	3.06	5.514279				
6	1j	3,4-di methyl	2-Br	4.21	5.375718				
7	1k	3,4-dichloro	4-F	6.27	5.202732				
8	11	3,4-dichloro	4-Cl	7.32	5.135489				
9	10	3,4-dichloro	4-OCH ₃	5.74	5.241088				
10	1t	3,4-dichloro	2-Br	6.43	5.191789				
11	1u	3,4-dibromo	4-F	8.92	5.049635				
12	1w	3,4-dibromo	4-Br	10.06	4.997402				
13	1x	3,4-dibromo	4-CH ₃	9.83	5.007446				
14	1y	3,4-dibromo	4-OCH ₃	8.09	5.092051				
15	2b	3,4-dibromo	2-F	13.37	4.874194				
16	2d	3,4-dibromo	2-Br	10.97	4.959793				
17	2f	3,4-dichloro	4-Cl	16.14	4.792096				
18	2i	3,4-dichloro	4-OCH ₃	17.19	4.74642				
19	2j	3,4-dichloro	2-Cl	24.15	4.617083				
20	2k	3,4-dichloro	4-H	28.69	4.542269				
21	21	3,4-dichloro	3,5-Dimethoxy	25.47	4.593971				
22	2m	3,4-dimethyl	4-F	15.56	4.80799				
23	20	3,4-dimethyl	4-Br	20.61	4.685922				
24	2q	3,4-dimethyl	4-OCH ₃	15.24	4.817015				
25	2r	3,4-dimethyl	2-CI	21.48	4.667966				

mostly determined by its ability to perform predictions of objects not included in the training sets. External validation of the QSAR model was carried out by examining the values of residuals, r²pred and r²m using test set compounds as shown in Table 7. Based on the external validation, r² predicted can be calculated which indicates the percent of predication of the selected model. Here outlier can be removed based on the value of the residue as shown in Figure 3

Prediction of novel compounds: For real validation, the biological activity of novel 35 compounds was predicted using equation 4. Among the predicted compounds, which exhibited potent activity will be synthesized. Table 8 shows a list of the predicted compounds having more predicted activity than observed activity in training and test set compounds.

and ketones (10 mmol, 1 equivalent) were dissolved in approximately 15 mL of ethanol. The mixture was allowed to stir for several minutes at 5–10 min. A 10 mL aliquot of a 40% aqueous potassium hydroxide solution was then slowly added dropwise to the reaction flask via a self-equalizing dropping funnel. The reaction solution was allowed to stir at room temperature for approximately 4 h. Most commonly, a precipitate formed and was then collected by suction filtration.¹⁸

b) (1A-1C) pyrazoline derivatives were prepared by taking the solution of chalcone in acetic acid in the presence of hydrazine hydrate. Reaction undergo cyclization and form final compound 1A-1C which is confirmed by characteristic peak.¹⁹

1-(5-(3,4-Dimethoxyphenyl)-3-(3-hydroxyphenyl)-4,5-dihydropyrazol-1yl) ethanone(1A)

Chemical

General procedure and Spectral Characterization

a) **(A-C)** Equimolar portions of the appropriately substituted aromatical dehydes (10 mmol, 1 equivalent)

Yield:	73.000	%; IR	(KBr,	in	cm ⁻¹):	3228.54(-0	DH),
2924.0	2(-OCI	H ₃), 17	42.18(-0	C=O), 1639	.31(-C=N	str.),
1416.0	6(-C-N	str.); ¹ I	H NMR	(DI	MSO, δ	in ppm): (5.67-
7.22(m	, 7H; .	Ar-H),	3.60(s,	3H;	3-OCH	H ₂), 3.23(s,	3H;

Table 2: Biological activities of test set compounds.										
Sr No.	Comp No.	<i>R</i> ,	R ₂	IC ₅₀	pIC ₅₀					
1	1d	3,4-di methyl	4-CH ₃	0.34	6.468521					
2	1i	3,4-di methyl	2-Cl	3.87	5.412289					
3	1m	3,4-dichloro	4-Br	6.85	5.164309					
4	1n	3,4-dichloro	4-CH ₃	7.38	5.131944					
5	1r	3,4-dichloro	2-F	6.56	5.183096					
6	1s	3,4-dichloro	2-Cl	7.28	5.137869					
7	1v	3,4-dibromo	4-Cl	8.15	5.088842					
8	2a	3,4-dibromo	4-NO ₂	10.144	4.973058					
9	2c	3,4-dibromo	2-Cl	12.46	4.907982					
10	2t	3,4-dimethyl	3,5-Dimethoxy	17.31	4.761703					

	Table 3: Correlation Matrix.												
	1	2	3	4	5	6	7	8	9	10	11	12	13
1	1												
2	0.207	1											
3	-0.410	-0.182	1										
4	-0.097	0.491	-0.236	1									
5	-0.149	0.348	0.315	0.2761	1								
6	0.141	-0.103	0.293	-0.286	-0.037	1							
7	0.317	0.270	0.159	0.053	-0.136	0.013	1						
8	0.050	0.170	0.180	0.378	0.476	0.034	0.167	1					
9	-0.205	0.039	-0.269	0.302	-0.398	-0.071	-0.110	0.0683	1				
10	0.151252	-0.149	0.243	-0.081	0.3178	-0.029	0.244	0.150	-0.462	1			
11	0.093	0.258	-0.356	0.411	0.054	-0.151	0.068	0.168	0.459	0.309	1		
12	-0.065	-0.039	0.361	0.027	0.130	-0.018	0.230	0.279	0.160	0.145	-0.066	1	
13	-0.463	0.109	0.268	0.374	0.412	-0.108	-0.150	0.209	-0.181	0.091	-0.071	0.357	1

OCH₃), 1.96(s, 3H; CH₃), 3.94,3.69(dd, 2H; CH₂), 4.59(s, ¹H; CH), 9.43(s, ¹H, OH); ¹³C NMR (DMSO, δ in ppm):151.7, 66.2, 39.9, 149.6, 147.8, 158.6, 135.4, 134.2, 109.8, 114.9, 121.9, 120.8,118.9, 118.2, 130.2, 164.5, 36.1, 56.1, 22.4; Exact mass: 340.14, Mass found : 341.14 (M⁺¹); Microanalysis: C, 67.05; H,5.92; N, 8.23; O, 18.80.

1-(5-(4-Chlorophenyl)-3-(4-ethoxyphenyl)-4,5dihydropyrazol-1-yl)ethanone (1B)

Yield : 75.44%; IR (KBr, in cm⁻¹): 3286.70 (-NH str.), 3057.47(-OC₂H₅), 1650.47(-C=O), 1598.92(-C=N str.), 1245.54(-C-N str.), 830.47(-Cl); ¹H NMR (DMSO, δ in ppm): 6.90-7.67(m, 8H; Ar-*H*), 1.41-1.45(t, 3H; CH₃), 4.04-4.09(q, 2H; CH₂), 3.055-3.11(dd, 1H; CH), 3.67-3.74(q, 1H; CH), 5.50-5.54(dd,1H; CH), 2.39(s, 3H; CH₃);¹C NMR (DMSO, δ in ppm): 151.7, 65.9, 39.9, 132.3, 161.7, 128, 139.1, 128.5, 114.5, 128.3, 127.3, 128.6, 114.5, 127.2, 128.3,168.5, 64.6,23.4, 14.8; Exact mass:342.11, Mass found : 343.12 (M⁺¹); Microanalysis: C, 66.57; H, 5.59; Cl, 10.34; N, 8.17; O, 9.33.

1-(5-(3,4-Dimethoxyphenyl)-3-(2-hydroxyphenyl)-4,5-dihydropyrazol-1-yl) ethanone (1C)

Yield: 66.05%; IR (KBr, in cm⁻¹): 3576.14 cm⁻¹ (-OH), 2935.46 cm⁻¹ (-OCH₃), 1654.98 cm⁻¹ (-C=O), 1516.67cm⁻¹ (-C=N str.) and 1256.02 cm⁻¹ (-C-N str.); ¹H NMR (DMSO, δ in ppm): 6.67-7.22(m, 7H; Ar-H), 3.60(s, 3H; 3-OCH₃), 3.23(s, 3H; OCH₃), 1.96(s, 3H; CH₃), 3.94,3.69(dd, 2H; CH₂), 4.59(s, 1H; CH), 9.73 (s, 1H, OH); ¹³C NMR (DMSO, δ in ppm): 151.7, 66.3, 40.2, 149.6, 147.8, 162.5, 118.6, 134.2, 109.8, 121.9, 117.8, 132.1, 118.9, 132.4, 121.4, 168.5, 56.1, 56.1, 23.4; Exact mass: 340.14, Mass found : 341.4 (M⁺¹); Microanalysis: C, 67.05; H, 5.92; N, 8.23; O, 18.80

Biological

SRB Assay

Synthesized compounds were evaluated for aromatase inhibiting activity in SRB (Sulphorhodamine B) assay. For screening experiments, cells were inoculated into 96 well microtiter plates in 90 µL at plating densities as shown in the study details above, depending on the doubling time of individual cell lines. After cell inoculation, the microtiter plates were incubated at 37° C, 5 % CO₂, 95 % air and 100 % relative humidity for 24 h before the addition of experimental drugs. After 24 h, one plate of each cell line was fixed *in situ* with TCA, to represent a measurement of the cell population for each cell line at the time of drug addition (Tz). Experimental drugs were solubilized in 0.1% DMSO at 400-fold the desired final maximum test concentration and stored frozen before use. At the time of drug addition, an aliquot of frozen concentrate was thawed and diluted to

10 times the desired final maximum test concentration with complete medium containing test article at a concentration of 10⁻³. Additional three, 10-fold serial dilutions were made to provide a total of four drug concentrations plus control. Aliquots of 10 µl of these different drug dilutions were added to the appropriate microtiter wells already containing 90 µl of the medium, resulting in the required final drug concentrations. After compound addition, plates were incubated at standard conditions for 48 hours and the assay was terminated by the addition of cold TCA. Cells were fixed *in situ* by the gentle addition of 50 μ l of cold 30 % (w/v) TCA (final concentration, 10 % TCA) and incubated for 60 min at 4°C. The supernatant was discarded; the plates were washed five times with tap water and air-dried. Sulforhodamine B (SRB) solution (50 μ l) at 0.4 % (w/v) in 1 % acetic acid was added to each of the wells and plates were incubated for 20 min at room temperature. After staining, the unbound dye was recovered and the residual dye was removed by washing five times with 1 % acetic acid. The plates were air-dried. The bound stain was subsequently eluted with a 10 mM trizma base and the absorbance was read on an Elisa plate reader

-	Table 4: List of QSAR equations.
Model No.	Coefficient of descriptor
1	CHIRAG-6788-03 = 5.08633 - 0.0562951 * Minimized_Energy + 0.00510767 * Molecular_Surface_Area_VAMP + 0.00145385 * PMI_Z + 2.48078 * RadOfGyration - 0.146542 * Shadow_XY
2	CHIRAG-6788-03 = 7.20669 - 0.0512013 * Minimized_Energy + 0.00149831 * PMI_Z + 2.61398 * RadOfGyration - 1.95471 * SIC - 0.140329 * Shadow_XY
3	CHIRAG-6788-03 = 7.55252 + 0.00558498 * Molecular_Surface_Area_VAMP + 0.000865382 * PMI_Z + 2.02687 * RadOfGyration - 4.96563 * SIC - 0.113323 * Shadow_XY
4	CHIRAG-6788-03 = 7.84899 + 0.00100139 * PMI_Z + 2.23756 * RadOfGyration - 4.28946 * SIC - 0.111674 * Shadow_XY + 0.00140674 * Strain_Energy
5	CHIRAG-6788-03 = 7.06335 + 0.0158803 * Molecular_Surface_Area_VAMP + 0.989815 * RadOfGyration - 8.72359 * SIC - 0.0804065 * Shadow_XY + 0.00292832 * Strain_Energy

Table	Table 5: Statistical parameters of the generated models.										
Model No	del No r ²		LOF	r ² _{pre}	r ² _m						
1	0.909937	0.8933	0.996889	0.996912391	0.608535						
2	0.913729	0.9231	0.998193	0.99777934	0.646335						
3	0.896199	0.9522	1.110648	0.998022999	0.637878						
4	0.884230	0.8832	1.10936	0.996912716	0.61252						
5	0.898102	0.8321	1.16331	0.99656207	0.584498						

at a wavelength of 540 nm with a 690 nm reference wavelength.(3)

Cell Culture

The compounds were synthesized and evaluated for their invitro anticancer activity using SRB Assay against MCF-7 cell lines using letrozole as reference compound. This study was taken to determine the half maximal inhibitory concentration (IC_{50}) of bioactive compounds. The cytotoxic activity also evaluated against healthy vero cell of compounds 1A, 1B, 1C, in order to express MCF-7 is a breast cancer cell line is the acronym of Michigan Cancer Foundation-7. MCF-7, cancer researchers can't obtain a mammary cell line that was capable of living longer than a few months. The cell lines were grown in RPMI (Roswell Park Memorial Institute) 1640 medium containing 10% fetal bovine serum and 2 mM L-glutamine.²¹

RESULTS AND DISCUSSION

For the preparation of targeted compound (1A-**1C**) synthesis was planned as per the given scheme in which substituted acetophenone is reacted with a substituted aldehyde in presence of FDC40% sodium hydroxide and Ethanol gave substituted chalcone which is further refluxed in presence of hydrazine hydrate and acetic acid to give the final product. After purification final compound was characterized by NMR spectroscopy, Mass spectrometry, IR spectroscopy. In IR spectroscopycompound A was confirmed at 1650 cm⁻¹ (conjugated diene) by the reported procedure.¹⁰⁻¹¹ The compound **1A** was confirmed by the disappearance of the peak at 1655cm⁻¹ indicating conjugated dienes and shows 1639.31(-C=N), 1416.06(-C-N str.) in IR spectroscopy, ¹H NMR (DMSO, δ in ppm): 6.67-7.22(m, 7H; Ar-H), 3.60(s, 3H; 3-OCH₂), 3.23(s, 3H; OCH₂), 1.96(s, 3H; CH₂), 4.59(s, 1H; CH), 9.43 (s, 1H, OH)in NMR spectroscopy, Molecular ion peak at $341.4(M^{+1})$ in Mass spectroscopy. Compound B was confirmed at 3407.54 (-NH str.), $2977.67(-OC_2H_z)$, 815.22(-Cl) by the reported procedure.¹⁰⁻¹¹ The compound 1B was confirmed by the disappearance of the peak at 1657cm⁻¹

indicating conjugated dienes and shows 1598.92(-C=N str.), 1245.54(-C-N str.), in IR spectroscopy, 3.67-3.74(q, 1H; CH), 5.50-5.54(dd,1H; CH), 2.39(s, 3H; CH₃) in NMR spectroscopy, molecular ion peak at 343.2 (M⁺¹) in Mass spectroscopy. Compound C was confirmed at 3434.26 (-OH), 2938.82(-OCH₃) by the reported procedure.¹⁰⁻¹¹ The compound **1B** was confirmed by the disappearance of the peak at 1655cm⁻¹ indicating conjugated dienes and shows3576.14(-OH), 2935.46(-OCH₃), 1516.67(-C=N), 1256.02(-C-N)in IR spectroscopy, 3.60(s, 3H; 3-OCH₃), 3.23(s, 3H; OCH₃),

Table 6: Results of the internal validation method for equation 4.								
Sr No.	Comp. No.	pIC ₅₀ (obs.)	pIC ₅₀ (pre.)	Residue				
1	1a	6.080922	5.833	-0.24792				
2	1b	5.866461	5.771	-0.09546				
3	1c	5.66354	5.975	0.31146				
4	1e	7.154902	6.464	-0.6909				
5	1g	5.514279	5.524	0.009721				
6	1j	5.375718	5.458	0.082282				
7	1k	5.202732	5.209	0.006268				
8	11	5.135489	5.205	0.069511				
9	10	5.241088	5.683	0.441912				
10	1t	5.191789	4.947	-0.24479				
11	1u	5.049635	5.044	-0.00564				
12	1w	4.997402	5.046	0.048598				
13	1x	5.007446	5.062	0.054554				
14	1y	5.092051	5.215	0.122949				
15	2b	4.874194	5.251	0.376806				
16	2d	4.959793	4.758	-0.20179				
17	2f	4.792096	4.574	-0.2181				
18	2i	4.74642	4.772	0.02558				
19	2ј	4.617083	4.534	-0.08308				
20	2k	4.542269	4.354	-0.18827				
21	21	4.593971	4.438	-0.15597				
22	2m	4.80799	4.754	-0.05399				
23	20	4.685922	4.659	-0.02692				
24	2q	4.817015	5.245	0.427985				
25	2r	4.667966	4.902	0.234034				

Table 7: Results of external validation using equation 4.									
Sr. No.	Comp. No.	pIC ₅₀ (obs.)	pIC₅₀(pre.)	Residue					
1	1i	5.412289	5.426	0.013711					
2	1m	5.164309	5.005	-0.15931					
3	1n	5.131944	5.009	-0.12294					
4	1r	5.183096	5.188	0.004904					
5	1v	5.088842	4.998	-0.09084					
6	2c	4.907982	4.976	0.068018					
7	2e	4.736364	4.792	0.055636					
8	2n	4.705093	4.655	-0.05009					
9	2р	4.75721	4.642	-0.11521					

	Table 8: List of the predicted compounds.								
Sr. No.	Compound No.	R ₁	R ₂	pIC ₅₀					
1	1A	4-Ethoxy	4-Chloro	7.858					
2	1B	2-Hydroxy	3,4-Dimethoxy	7.764					
3	1C	3-Hydroxy	3,4-Dimethoxy	7.449					

Table 9: Percent Cell growth inhibition data on MCF-7 cell line.									
Concentration (µg/ml)	Log Conc.	% Inhibition of Cell Growth							
		Std.	1A	1B	1C				
0.05	-1.29	-27.31	-30.11	-28.39	-31.22				
0.15	-0.82	-21.25	-28.44	-25.88	-30.89				
0.46	-0.34	-16.87	-22.11	19.34	-23.59				
1.37	0.14	-10.89	-13.55	-10.12	-15.39				
4.12	0.61	-4.84	-11.24	-8.63	-11.44				
12.35	1.09	3.36	6.55	15.55	7.83				
37.04	1.57	10.47	18.63	21.26	15.76				
111.11	2.05	17.32	25.96	29.64	22.64				
333.33	2.52	24.00	30.34	36.74	32.59				
1000.00	3.00	32.00	31.27	39.14	33.24				





Figure 1: Schematic diagram for the synthesis of pyrazoline derivatives



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Table 10: IC ₅₀ value of synthesized compound.										
Compound	PIC 50pre	IC _{50pre} μΜ	PIC 50obs	IC _{50obs} μM	Residual	CC _{₅₀} Vero cell	SI			
1A	7.858	0.013	7.224	0.059 ±0.019	-0.63	85.12	≥1000			
1B	7.764	0.017	7.982	0.010 ±0.011	0.218	72.78	≥1000			
1C	7.449	0.031	7.425	0.037 ±0.013	-0.024	77.35	≥1000			
Letrozole				0.025±0.01		89.47	≥1000			



Figure 3: Graph of observed and predicted activities of the training set compounds by Equation 4.

3.02(s, 3H; CH₂), 1.14 (dd, 2H; CH₂), 2.59(s, 1H; CH), 9.73 (s, 1H, OH) in NMR spectroscopy, Molecular ion peak at 341.4 (M⁺¹) in Mass spectroscopy. All of the synthesized compounds (1A,1B,1C) were evaluated for their aromatase inhibiting activity as shown in Table 9, activity was taken on MCF-7 cell line through SRB assay in which IC₅₀ was calculated independently for every compound as shown in Table 10. Among all the synthesized compound 1B showing better anticancer activity having IC50 value 0.010 µMas compared to standard drug letrozole, compound 1B, compound 1C. compound 1C is having two methoxy group at 3rd and 4th position of an aromatic ring which is bulkier and sterically hindered compare to compound 1B and compound 1A hence it is least active, compound 1A is also having two methoxy group but are less sterically hindered than compound 1C, compound 1A having only one electron-withdrawing group chloro at the para position of the aromatic ring which may be the possible reason for the significant activity.

CONCLUSION

The research work illustrates the attention to novel pyrazoline derivatives (**1A-1C**) representing anticancer agents through aromatase enzyme inhibition which involve in estrogen biosynthesis in breast cancer. The development of novel anticancer agent for the treatment of breast cancer through aromatase inhibition concludes that there is increasing need to develop novel derivatives which can inhibit aromatase enzyme which is crucial for estrogen synthesis in breast cancer.

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CONFLICT OF INTEREST

Authors declare no conflict of interest in this research work.

ABBREVIATIONS

NMR: Nuclear magnetic resonance spectroscopy; TMS: Tetramethylselane; DMSO: Deuterated dimethyl sulfoxide; LC-MS: Liquid Chromatography-Mass Spectrometry; FT-IR: Fourier Transform infrared spectroscopy; SRB: Sulforhodamine B; QSAR: Quantitative Structure-activity relationship; TLC: Thin layer Chromatography.

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SUMMARY

• Being good pharmacophorepyrazolinefor anticancer series waswidely used for their potential in the present study.

• Some new 1-(5-(4-Chlorophenyl)-3-(4-ethoxyphenyl)-4,5-dihydropyrazol-1-yl) ethanone potential against cancer cell line MCF-7activity having IC50value 0.010 μM.

• QSAR study of pyrazoline as anticancer moiety was performed using Accelrys discovery studio client (Version 2.1) as the modelling tool. A total of 35 selected molecules were considered for the development of the QSAR model.

• The statistical analysis performed revealed following observations; r2 = 0.884, r2cv (Cross validated r2) = 0.883, r2pred = 0.99 and r2m = 0.61. A total of 13 descriptors pruned on the study explained the structural correlation of pyrazoline.

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