

Solid State Characterization and Tableting Studies of Ethanol Based Cocrystals of Fenofibrate with Nicotinamide

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ABSTRACT

Objectives: The pharmaceutical cocrystals can be defined as dissociable “API-excipient” molecular complexes or Co-crystals are solids that are crystalline materials composed of two or more molecules in the same crystal lattice as per U.S. Department of Health and Human Services Food and Drug Administration Center for Drug Evaluation and Research (CDER) April 2013. The objectives of present investigation were to formulate cocrystals of fenofibrate with nicotinamide by solution cocrystallization technique. **Methods:** The cocrystals were prepared by solution cocrystallization technique by using ethanol as a solvent and nicotinamide as a cofomer in the ratio of 1:1 & 1:2. The prepared cocrystals were evaluate for, solid state characterization by FTIR, PXRD, Raman spectroscopy and evaluated for tableting performance. **Results:** The formation of cocrystals has been confirmed by FTIR and Raman spectroscopy. The formulations 1:1 and 1:2 ratios showed better flow properties. The order of tableting performance 1:2 ratio cocrystals > 1:1 ratio cocrystals > Recrystallized nicotinamide > Nicotinamide >> Recrystallized Fenofibrate >> Fenofibrate. **Conclusion:** The cocrystals showed superior tableting performance. This might be due to the co-crystals, contains hydrogen-bonded two dimensional flat slip planes which exhibits higher plasticity.

Key words: Cocrystals, Crystallization, Fenofibrate, Nicotinamide, Tableting

INTRODUCTION

Cocrystals can be defined as crystalline materials consist of two or more different components (or commonly called multi-component crystals). For the pharmaceutical cocrystals, one component is an active pharmaceutical ingredient and other components are called as cofomers. Cocrystals have gained considerable interest in pharmaceutical research due to its ability to improve physicochemical characteristics of an API such as such as mechanical behavior, compressibility, solubility, dissolution rate, moisture stability, and bioavailability of drugs with their chemical structure unchanged.^{1,2} In a short time span cocrystals are of the interest

of the researchers because fast forward to 2015 and the first example of a pharmaceutical ionic cocrystal drug was approved by the US-FDA in accordance with its cocrystal guidance paper for drug substances in 2012 – 2013.³ Entresto is a novel drug launched by Novartis for the treatment of chronic heart failure with the composition monosodium sacubitril, disodium valsartan, and water. Other examples include escitalopram oxalate.⁴ a marketed drug in a crystal form that is composed of protonated escitalopram cations, water molecules, oxalate dianions, and diprotonated oxalic acid molecules. The anti-diabetes drug Ertugliflozin.⁵ currently

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in phase III clinical development is a cocrystal with 5-oxo-proline.

Fenofibrate is a neutral, lipophilic compound that is practically insoluble in water, making it challenging to consistently achieve therapeutic levels.⁶ Our previous investigation showed nicotinamide forms stable cocrystals theoretically and practically with fenofibrate with improved dissolution rate.⁷

In general, fenofibrate products can be grouped with regard to their food effects. For example, fenofibrate formulations that should be taken with meals include non-micronized tablets (Fenoglide[®], Lofibra[®], generic), micronized capsules (Lofibra[®] and generic), micro coated micronized tablets (Lofibra[®]), and fenofibrate hard gelatin capsules (Lipofen[®]). Fenofibrate formulations that can be taken with or without meals include nanoparticle tablets (Tricor[®]), IDD-P tablets (Triglide[®]), micronized capsules (Antara[®]), and the choline salt of fenofibric acid (Trilipix[®]) in a dosage range of 40 to 200 mg.⁸

Literature reveals that Fenofibrate (FNO) shows poor flowability and compaction properties. To enhance the compressibility spherical crystallization, highly plastic granules,⁹ and melt sonication techniques has been reported.¹⁰

Cocrystallization of the drug powder to control the crystal packing and modulate tableability which describes the capacity of a powder to be transformed into a tablet in a broader manufacturing perspective and compaction which enables bonding between particles are important improvement at the preformulation stage. The objectives of present investigation were to formulate cocrystals of fenofibrate with nicotinamide by solution cocrystallization technique, solid state characterization of cocrystals by FTIR, PXRD and Raman spectroscopy and to evaluate cocrystals for tableting performance.

MATERIALS AND METHODS

Materials

Fenofibrate was obtained as a gift sample from Alembic pharma, Pvt.Ltd. Vadodara. Nicotinamide were procured from Loba Chemie Pvt. Ltd. Mumbai. All other chemicals were of analytical grade.

Preparation of Cocrystals by Solution crystallization technique

Fenofibrate and nicotinamide were weighed in 1:1 molar ratio, dissolved in 20 ml of ethanol with sonication. The saturated solution was kept overnight to evaporate solvent. The crystals were obtained after evaporation of ethanol; crystals were allowed to dry in the air.³

Fourier Transform Infrared spectroscopy (FTIR)

IR spectroscopy was conducted using a Shimadzu FTIRAffinity-1S Spectrophotometer (Shimadzu) and the spectrum was recorded in the wavelength region of 7800-350 cm^{-1} . The procedure consisted of dispersing a sample (drug alone, or prepared cocrystals) in KBR.

Powder X-Ray Diffraction Studies (P-XRD)

The X-ray diffraction patterns of pure drug and the optimized crystals formulation were recorded using Bruker analytical X-ray diffraction (D2 Phase, Bruker) (Germany) with a copper target over the interval of 5-700 $2\theta^{-1}$. The conditions were voltage 20kV, current 30 mA, scanning speed 20 /min, temperature of acquisition: room temperature, detector: scintillation counter detector, sample holder: non-rotating holder.¹

Raman Spectroscopy

The Raman spectra of the pure drug and prepared formulations were recorded in the spectral range of 0–5000 cm^{-1} using a Raman spectrometer (Bruker MultiRAM, Germany Make) equipped with an Nd: YAG laser source with an at excitation wavelength of 1064 nm and resolution 4 cm^{-1} .

Tableting Studies

Approximately 250 mg of powder was manually filled into a KBR press die and compressed at predetermined pressures (1 to 2.5 tons) to get flat-faced round tablets (8 mm) diameter, and thickness was measured by using digital venire caliper. The hardness was measured by using Monsanto Hardness Tester. The prepared formulations were also evaluated for Micromeritic properties

RESULTS AND DISCUSSION

Fourier Transform Infrared spectroscopy (FTIR)

The result of FTIR revealed considerable changes in the IR peaks of fenofibrate in prepared co-crystals when compared to pure drug thereby indicating the presence of hydrogen bonding in the co-crystals as shown in Table 1. Specific FNO peaks were observed at 2982 cm^{-1} indicates aromatic C-H stretching, peak at 1587 cm^{-1} indicates C=O stretching whereas, peaks at 1241 cm^{-1} and 1087 cm^{-1} indicates aralkyl and dialkyl ether C-O stretching respectively. Also, peak at 757 cm^{-1} indicates presence of halogen-hydrogen interaction. Specific nicotinamide peaks are also observed at 3151.69 cm^{-1} , 1283 cm^{-1} and 1726.29 cm^{-1} indicating presence of aromatic –NH₂ group, CN- stretching and –C=O stretching respectively. In case of co-crystals prepared by 1:1 F+N and 1:2 F+N methods induces interaction which is

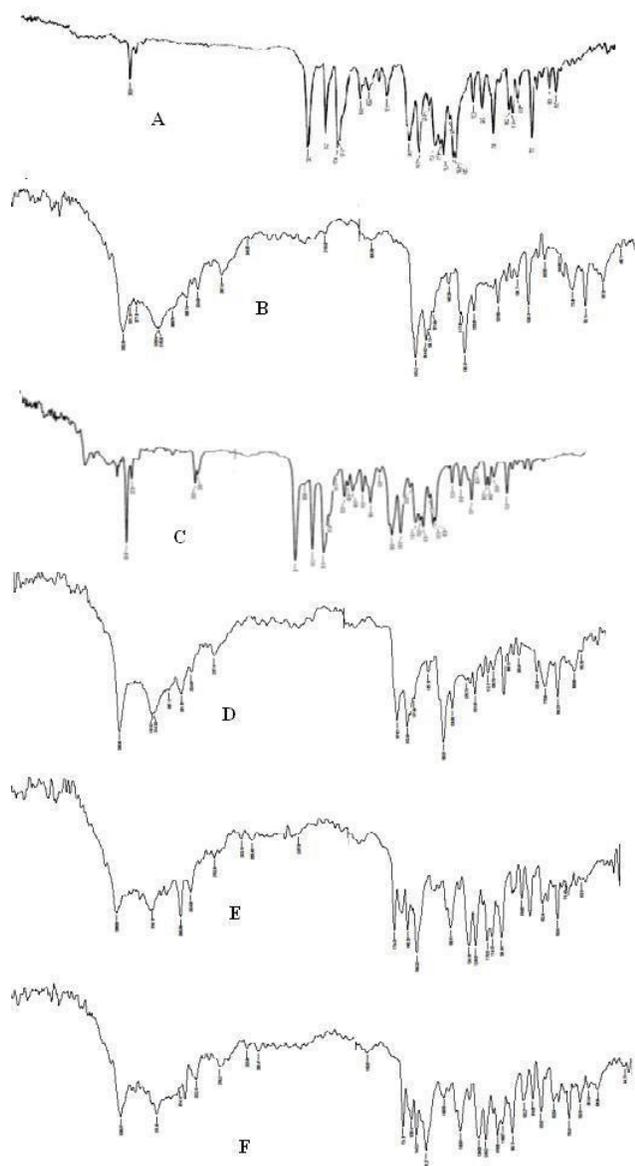


Figure 1: FTIR spectra's A. Pure fenofibrate B. Recrystallized Fenofibrate C. Pure Nicotinamide D. Recrystallized Nicotinamide E. 1:1 ratio cocrystals F. 1:2 ratio cocrystals.

indicated by appearance of characteristics peak of aromatic -NH_2 group at 3151.69 cm^{-1} that might be involved in the interaction as shown in Figure 1. Those phenomena may be caused by the hydrogen bond interactions involved in co-crystal formation which changes the symmetry characteristic.

Powder X-Ray Diffraction Studies (P-XRD)

PXRD of all powders showed intense sharp diffraction peaks without amorphous halo observed. The PXRD pattern of fenofibrate as shown in Figure 2 indicates crystalline nature of drug. The sharp intense peaks at 2θ - 11.88° , 16.13° , 20.75° , 22.17° , 24.55° showed that the

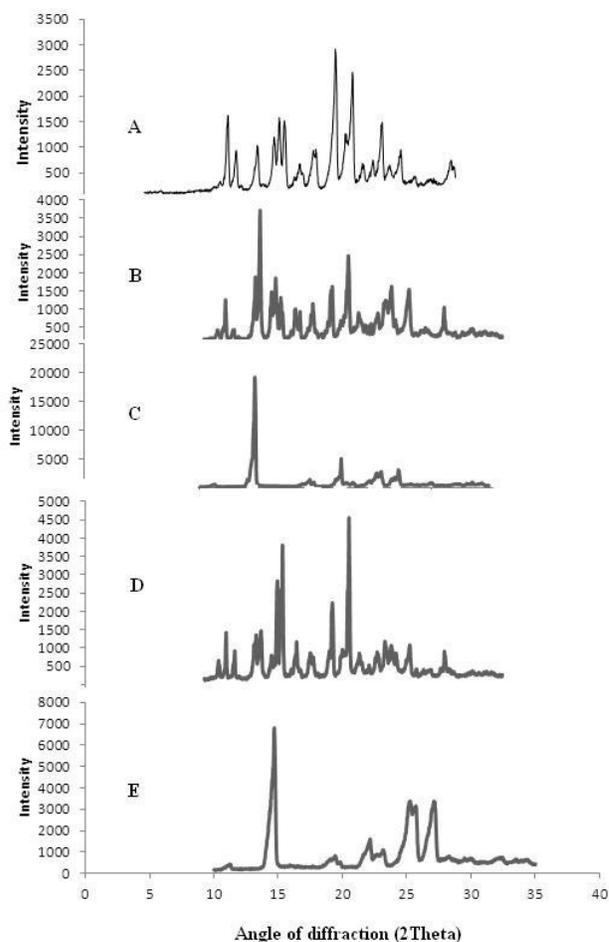


Figure 2: PXRD spectra's A. Pure fenofibrate B. Recrystallized Fenofibrate C. Pure Nicotinamide D. 1:1 ratio cocrystals E. 1:2 ratio cocrystals.

sample was fenofibrate. The pure nicotinamide showed intense peak at diffraction angle of 14.73° and 22.20° . The PXRD pattern of 1:1 ratio cocrystals showed shifting in 2θ to 11.82° , 14.73° , 20.68° , 22.12° , 25.67° and 1:2 ratio showed intense peak at 2θ - 14.67° , 22.04° , 25.14° , 27.05° as shown in Figure 2. This revealed a characteristic diffraction pattern, which differed from those of the two individual components. Hence shifting of the peaks has been observed in formulations. This shifting is a clear indication of formation of new phase's i.e. cocrystals.^{11,12} as confirmed from FTIR.

Raman Spectroscopy

The fenofibrate showed Raman bands at 1600 cm^{-1} and 1650 cm^{-1} for aromatic chain vibrations, $\text{C}=\text{O}$ functional group respectively. The $\text{C}=\text{C}$ bond is always strong and very reliable. "To form a salt with amine bases the carbonyl group bands shift to lower side by $30\text{-}40\text{ cm}^{-1}$ ". Such type of shifting has not been observed with formulations so it confirms the formed phases are

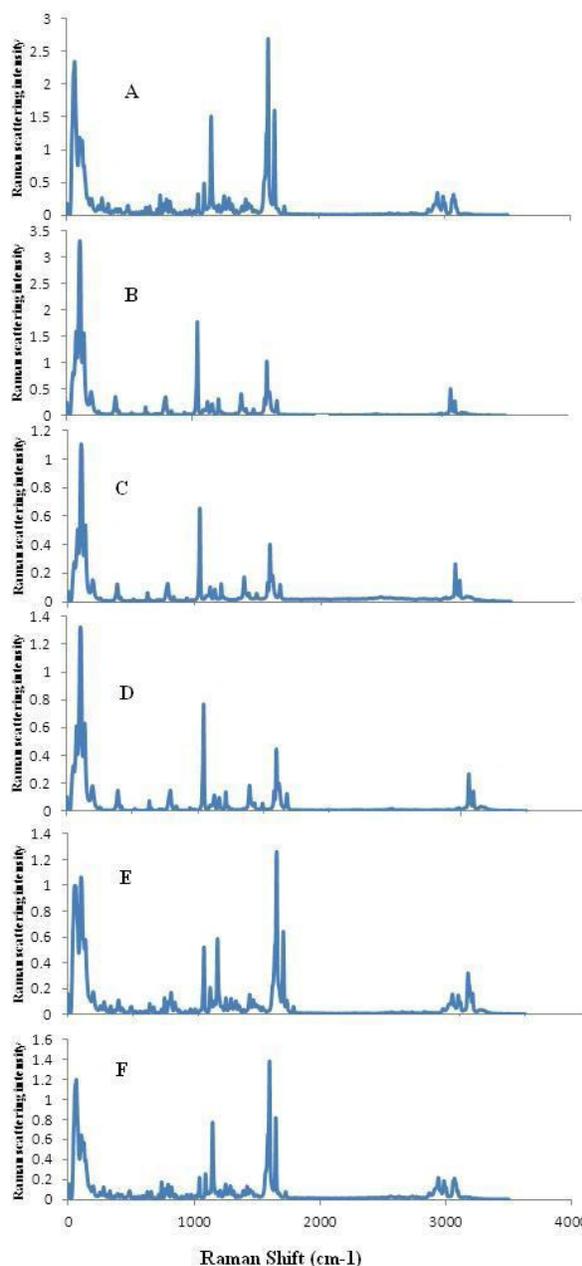


Figure 3: Raman Spectra's A. Pure Fenofibrate B. Recrystallized Fenofibrate C. Pure Nicotinamide D. 1:1 ratio cocrystals E. 1:2 ratio cocrystals.

cocrystals. The C-H stretching region in between (3000-2850 cm^{-1}) is a very good diagnostic bond for the presence of the aldehyde functional group.^{13,14,15} Similar to FTIR, some peak shift also take place for Raman spectra which can be attributed to intermolecular hydrogen-bonding. The results are mentioned in Table 2 and Figure 3.

Tableting Studies

The Table 3 shows among all the formulations the 1:1 and 1:2 ratios of formulations showed better micromeritic

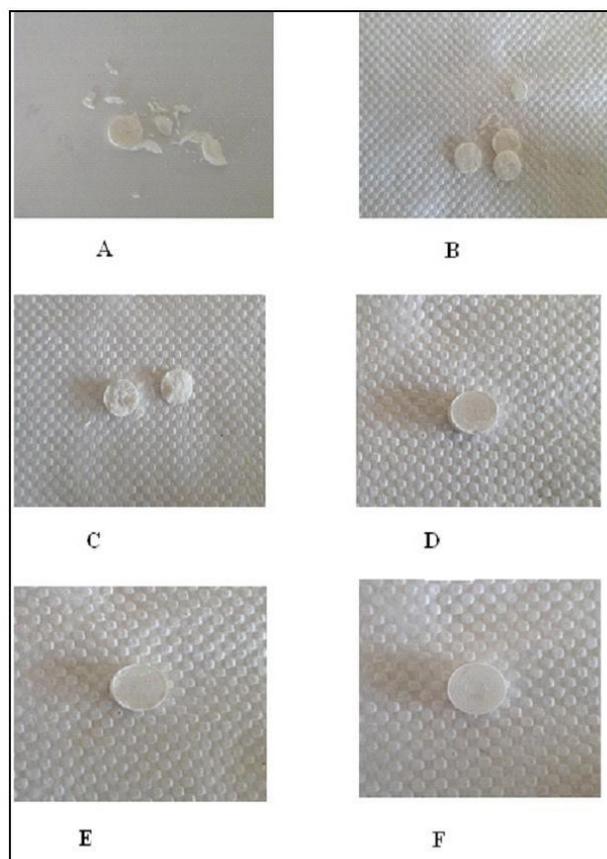


Figure 4: Photographs of tablets compressed at a pressure and weight (1.0 ton and 250 mg) respectively.

properties. The pure fenofibrate showed good angle of repose but moderate type of Hausner's ratio 1.399 and the having poor Carr's index 28.52 %.^{16,17}

The hardness and thickness of tablet at different applied pressure was determined. The 1:2 ratio based cocrystals are very useful in the formulation of stable tablet than pure drug. The 1:2 ratios of cocrystals of fenofibrate with nicotinamide showed better thickness and hardness properties among those formulations. The order of tableting performance 1:2 ratio cocrystals > 1:1 ratio cocrystals > Recrystallized nicotinamide > Nicotinamide >> Recrystallized Fenofibrate >> Fenofibrate as shown in Table 4 and 5. At any given compaction pressure, the tablet tensile strength of the co-crystal was always greater than that of pure fenofibrate and nicotinamide. The different mechanical properties of Fenofibrate, Nicotinamide, and their 1: 1 co-crystal originate from differences in crystal packing. Fenofibrate crystals, which contain three dimensional hydrogen bonded networks, may exhibit low plasticity and high resistance to deformation.¹⁸

Table 1: Summary of FTIR Peaks of Different Formulations

| Formulation s | C=O Str. | C=C | C=O Aryl esters | H-X | CN | NH ₂ |
|--------------------------------------|----------|---------|-----------------|--------|---------|-----------------|
| Fenofibrate (A) | 1587 | 1647.21 | 1726.29 | 757 | - | - |
| Crystallized Fenofibrate (B) | 1587.06 | 1647.21 | 1724.36 | 763.81 | - | - |
| Nicotinamide (C) | 1571.99 | 1683.86 | - | 763.81 | 1288.52 | - |
| Crystallized Nicotinamide (D) | 1571.99 | 1674.21 | - | 776.30 | 1263.76 | 3067.17 |
| Cocrystals (1:1 ratio) (E) | 1595.13 | 1645.34 | 1726.29 | 763.81 | 1283.52 | 3047 |
| Cocrystals (1:2 ratio) (F) | 1597.66 | 1645.39 | 1726.29 | 763.81 | 1283 | 3151.69 |

Table 2: FT RAMAN Ranges of Different Formulations

| Functional Group | Lattice vibrations | u(C=S) | u(CC)aromatic ring chain vibrations | u(C=C) | u(C-H) | u(C=H) |
|-------------------------------|--------------------|---------|-------------------------------------|---------|---------|--------|
| Fenofibrate (A) | 64.99 | 1146.9 | 1600 | 1650 | 2990.56 | - |
| Crystallized Fenofibrate (B) | 107.41 | 1042.7 | 1598.16 | - | - | 3063.8 |
| Nicotinamide (C) | 109.34 | 1042.7 | 1598.16 | - | - | 3063.8 |
| Crystallized Nicotinamide (D) | 109.34 | 1042.7 | 1598.17 | - | - | 3063.8 |
| Cocrystals (1:1 ratio) (E) | 111.27 | 1044.6 | - | 1650.23 | 2990.60 | 3065.7 |
| Cocrystals (1:2 ratio) (F) | 64.99 | 1146.89 | - | 1650.23 | 2986.70 | 3073.8 |

Table 3: Micromeritic Properties of Formulations

| Formulation code | Angle of repose | Bulk density (gm/cm ³) | Tapped density (gm/cm ³) | Carr's Index (%) | Hausner's ratio |
|-------------------------------|-----------------|------------------------------------|--------------------------------------|------------------|-----------------|
| Fenofibrate (A) | 26.51±1.25 | 0.238±0.012 | 0.333±0.013 | 28.52±1.20 | 1.399±0.13 |
| Crystallized Fenofibrate (B) | 31.54±1.41 | 0.274±0.021 | 0.333±0.015 | 17.71±1.32 | 1.21±0.21 |
| Nicotinamide (C) | 26.51±1.75 | 0.274±0.31 | 0.320±0.028 | 14.375±1.52 | 1.167±0.24 |
| Crystallized Nicotinamide (D) | 25.032±1.20 | 0.245±0.023 | 0.316±0.023 | 22.46±2.02 | 1.28±0.26 |
| Cocrystals (1:1 ratio) (E) | 29.42±1.34 | 0.304±0.024 | 0.337±0.027 | 9.55±1.52 | 1.10±0.32 |
| Cocrystals (1:2 ratio) (F) | 28.94±1.46 | 0.297±0.028 | 0.320±0.025 | 7.0±1.23 | 1.07±0.04 |

Table 4: Thickness (Mm) of Tablet at Deferent Pressure

| Compression Pressure in (tons) | A | B | C | D | E | F |
|--------------------------------|---|---|-----------|-----------|-----------|-----------|
| 1 | - | - | - | 3.65±0.04 | 3.44±0.02 | 3.60±0.02 |
| 1.5 | - | - | - | 3.40±0.04 | 3.34±0.02 | 3.45±0.02 |
| 2 | - | - | 3.13±0.04 | 2.90±0.04 | 2.95±0.02 | 2.88±0.02 |
| 2.5 | - | - | 3.00±0.04 | 2.88±0.04 | 2.85±0.02 | 2.82±0.02 |

Table 5: Hardness (Kg/Cm²) Of Tablet at Different Pressure (Mean Values)

| Compression Pressure in (tons) | A | B | C | D | E | F |
|--------------------------------|---|---|------|------|------|------|
| 1 | - | - | - | 3.00 | 3.10 | 3.30 |
| 1.5 | - | - | - | 3.10 | 3.30 | 3.50 |
| 2 | - | - | 2.89 | 3.68 | 3.64 | 3.70 |
| 2.5 | - | - | 2.99 | 3.69 | 3.74 | 3.79 |

CONCLUSION

The cocrystals of antihyperlipidemic drug, fenofibrate with coformer nicotinamide have been successfully prepared. Cocrystals gives reliable thickness and hardness of the tablets. The cocrystal displayed superior tableting performance. This might be due to the co-crystals, containing hydrogen-bonded two dimensional flat slip planes exhibit higher plasticity.

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CONFLICT OF INTEREST

The authors report no conflicts of interest. The authors alone are responsible for the content and writing of the paper.

ABBREVIATIONS USED

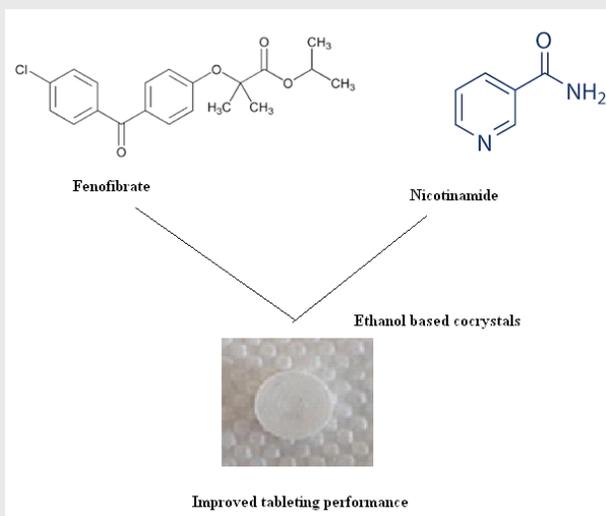
FNO: Fenofibrate.

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PICTORIAL ABSTRACT



SUMMARY

- Fenofibrate forms the cocrystals with Nicotinamide which has been confirmed by instrumental methods of analysis and the order of tableting performance 1:2 ratio cocrystals > 1:1 ratio cocrystals > Recrystallized nicotinamide > Nicotinamide >> Recrystallized Fenofibrate >> Fenofibrate. This might be due to the co-crystals, containing hydrogen-bonded two dimensional flat slip planes exhibit higher plasticity.

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