# Formulation and Evaluation of Gastro-bilayer Floating Tablets of Ezetimibe as Immediate Release Layer and Atenolol as Sustained Release Layer

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# ABSTRACT

Introduction: Combination therapy of ezetimibe and atenolol is highly desirable for better management of dyslipidaemia and hypertension. Ezetimibe has poor solubility hence variable and low bioavailability. Atenolol has poor absorption in lower gastrointestinal tract, short half-life. Therefore, the present study was to develop gastro-bilayer floating matrix tablet in which ezetimibe was incorporated as immediate layer and atenolol as sustained release layer. Methods: Solubility of the ezetimibe was enhanced by solid dispersion technique and was characterized by FTIR, DSC and XRD study. Gastrobilayer floating tablets were prepared by direct compression method. Results and Discussion: Hydroxypropyl methylcellulose K100 (37.5 % w/w) as release retardant and croscarmellose sodium (15 % w/w) as superdisintegrants in immediate layer as optimized. The total floating time of the optimized tablet was 12 h with 9 min of floating lag time. Atenolol release was sustained through diffusion mechanism over 12 h and more than 95 % ezetimibe was released within 30 min. Conclusion: It can be concluded that biphasic drug release pattern was successfully achieved through the formulation of gastro-floating bilayer tablets in this study, allowing strengthened combination therapy for hypertension and dyslipidemia.

Key words: Gastro-bilayer, floating, immediate release, sustained release, ezetimibe, atenolol.

# INTRODUCTION

Frequent dosing in oral drug delivery results in fluctuation of plasma drug concentration and finally toxicity.<sup>1</sup>To overcome this problem oral controlled drug delivery system was developed that deliver the drug for an extended period of time. Physiological problems like drugs with narrow absorption window, alteration in emptying time of stomach, drugs that has stability issues in intestine and drugs that are transported via active transport mechanism<sup>2</sup> were the difficulties of this system. To overcome these difficulties gastroretentive drug delivery system (GDDS) has been developed. Gastroretentive systems can remain in the gastric region for several hours and prolong the gastric residence time of drugs. Prolonged gastric retention improves bioavailability,

reduces drug waste and improves solubility for drugs that are less soluble in a high pH environment. It also provides local drug delivery to the stomach and proximal small intestine.<sup>3</sup> Among the gastroretentive dosage forms, floating drug delivery system is considered to be most favourable because it does not intensely affect the motility of GIT.4 The matrix bilayer tablet with two separate release-layers is a biphasic delivery system that aims to deliver drug at two different rates or simultaneously releases two drugs with the benefits of formulating two chemically incompatible drugs into a system, simultaneously releasing two APIs with desired release profiles, increasing efficacy of API by a synergistic effect,

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decreasing the dosing unit burden and better patient compliance.<sup>5</sup>

Cardiovascular disease is a collective term that covers any disease of heart and circulatory system.<sup>6</sup> The foremost risk factors for cardiovascular diseases are tobacco exposure, hypertension, high cholesterol, alcohol consumption, obesity. Globally cardiovascular diseases accounts for approximately 17 million deaths a year of these 9.4 million deaths are due to hypertension.<sup>7</sup> The first National Health and Nutritional Examination Survey (NHANES) estimated the co-prevalence of hypertension and elevated total cholesterol. During the third NHANES, around 60 % of adults had concomitant hypercholesterolemia and 55 % of the adults with hypercholesterolemia had hypertension.8,9 Combination therapies that target several risk factors, such as hypercholesterolemia, hypertension, diabetes and artery function may more effectively treat cardiovascular disease than a therapy that targets several risk factors for the cardiovascular death.<sup>10-12</sup> So, combination therapy is required to deliver drugs for dyslipidaemia and hypertension.

Atenolol is a cardio-selective  $\beta$ -1 adrenoceptor devoid of intrinsic sympathomimetic and membrane stabilizing activity. It belongs to anti-hypertensive category and therefore reduces the high blood pressure.13 It has absorption window in upper GIT and the poor absorption in lower GIT. This varied absorption results in lowering bioavailability i.e., 50 % with a half-life of 6-7 h.14 Thus, it seems that increase in gastric residence time may increase the extent of absorption and bioavailability of drug. Hence, it was selected as sustained release layer to improve bioavailability. Ezetimibe is classified under anti-hyperlipidemic class and which is a BCS class II drug.<sup>15</sup> So, the solubility must be improved. It has longer biological half-life (22 h) and its bioavailability was 30-60 %.16 So, it can be formulated into an immediate release layer to achieve better absorption. The poor level of control of hypertension and dyslipidemia highlights the need for new strategies to manage these risk factors thereby reducing the impact of cardiovascular diseases. A single pill combination of an antihypertensive and lipid lowering medication may address some of the issues through to hinder the management of cardiovascular diseases, such a poor adherence to multiple treatments due to high pill burden and the reluctance of physicians to manage more than one of cardiovascular risk factors simultaneously. Swain et al.17 and Kulkarni et al.18 worked on development and evaluation of bilayer floating tablets of atenolol and simvastatin/lovastatin for biphasic release profile to improve the bioavailability of atenolol and to treat

hyperlipidemia which is the major problem in hypertensive patients.<sup>4</sup> Ezetimibe is a lipid - lowering agent acts immediately and reduces the cholesterol absorption and then atenolol, an antihypertensive agent sustains its action over longer period of time which may useful for management of cardiovascular diseases. To the best of our knowledge, no reported literature was found on the combination therapy of ezetimibe and atenolol. In the present study we have made an effort to formulate a gastro-bilayer floating system of ezetimibe in the immediate release layer and atenolol in the sustained release layer to improve their bioavailability.

# MATERIALS AND METHODS

#### Materials

Atenolol was procured from Yarrow chem products (Mumbai, India). Ezetimibe was a gift sample from Aurobindo Pharma Pvt. Ltd., India. Hydroxy propyl methyl cellulose, PVP K- 30, PEGs, croscarmellose sodium were supplied by Yarrow chem. Pvt. Ltd., Mumbai. Sodium bicarbonate and lactose were purchased from Thermo Fischer Scientific Pvt. Ltd. (Ahmedabad, India). Microcrystalline cellulose was obtained from Himedia (Gurgaon, India). Ethanol and HCl were from Fischer Scientifica, India and of analytical grade.

#### Methods

#### Solubility enhancement of Ezetimibe

Solubility of ezetimibe was enhanced by preparing solid dispersions (SDs) by solvent evaporation method (SM) using PVP K-30 (ESD1), PEG 6000 (ESD2) and PEG 8000 (ESD3) in the ratio of 1:1 w/w (drug: carrier). Ethanol (10 mL) used as solvent to solubilize the ezetimibe and polymers by continuous stirring with a magnetic stirrer (RemiElektrotechnik Limited, India, Model: 1MLH) for an hour at room temperature. The mixture was stored at a room temperature for complete evaporation of ethanol. The resulting SDs were scraped, pulverised and passed through 40# mesh sieve. The samples were stored in screw cap glass vials and kept in desiccators until further analysis.<sup>19,20</sup>

#### **Characterization of SDs**

# FTIR spectroscopy

FTIR spectra of ezetimibe, PVP K-30 and SD were subjected to compatibility studies by diamond ATR spectrophotometer (Cary 60, Agilent technologies, Germany). The prepared SDs were scanned at a resolution of 2 cm<sup>-1</sup>, from 4000 to 400 cm<sup>-1</sup>.

#### Differential scanning calorimetry (DSC)

Thermal analysis was carried out for ezetimibe, PVP K-30 and SD using DSC (Pyris Diamond, Singapore). The samples were heated at a constant rate of 10°C/min over a temperature range of 30-250°C using platinum crucible with alpha alumina powder as reference to predict any physicochemical interactions between components.

# X-ray diffraction analysis (XRD)

Ezetimibe, PVP K-30 and SD were subjected to X-ray diffraction analysis, using Cu target slit 10 mm (ULTIMA III, Japan) to investigate physical state of SD.

#### In vitro dissolution studies

Dissolution studies were performed for the SD using USP dissolution type II apparatus (paddle method) with 50 rpm in 900 ml of 0.1 M HCl as dissolution medium at  $37 \pm 0.5^{\circ}$ C. The samples were withdrawn at predetermined time intervals (5, 10, 15, 30, 45 and 60 min) and the same amount of preheated ( $37 \pm 0.5^{\circ}$ C) fresh medium was added to maintain constant volume throughout study. The percentage drug release values obtained from the dissolution studies were plotted against time.

#### Compression of immediate release tablets

SDs equivalent to 10 mg of ezetimibe was blended with croscarmelose sodium and lactose to prepare IR1, IR2, IR3 formulations and passed through sieve 44# then added magnesium stearate before direct compression of tablets (Table 1).

# Compression of sustained release tablets

All the ingredients of formulations (SR1-SR9) shown in Table 2 were accurately weighed and passed through sieve 44# to obtain uniform size. Magnesium stearate

Table 1: Formulation of immediate release layer     tablets by direct compression method							
Formulation Code SD equivalent to (ESD1)		CCS (mg)	Lactose (mg)	Magnesium Stearate (mg)	Total (mg)		
IR1	40	5	54	1	100		
IR2	40	10	49	1	100		
IR3	40	15	44	1	100		

was added before punching of tablets by direct compression method.

# Preparation of gastro-bilayer floating tablets

Immediate release tablets (IR3) and sustained release tablets (SR9) were selected based on dissolution profile. The gastro-bilayer floating tablets were prepared with varying concentration of sodium bicarbonate (3.5, 5.0, 6.25, 7.5 % w/w) in the sustained release layer (Table 3). Bilayer floating tablets were formulated with multistation rotary tablet punching machine via single compaction method. Initially, sustained release layer powder blend was accurately weighed and fed into die cavity of tablet punching machine. Similarly, immediate release layer powder was fed to die cavity which was previously filled with the sustained release layer and compressed so that the final hardness obtained for the bilayer floating tablet was between 5.5 -  $6.5 \text{ kg/cm}^2 \text{ using } 12 \text{ mm}$  flat punches.

# Post compression parameters of gastro-bilayer floating tablets

#### Weight variation

Twenty tablets were randomly selected from each batch and calculated the percentage deviation of individual tablet weight from average weight of tablets.

Table 2: Formulation of sustained release layer by direct compression method.									
Formulation Code	Drug (mg)	HPMC K4 (mg)	HPMC K15 (mg)	HPMC K100 (mg)	MCC (mg)	Magnesium Stearate (mg)	Total (mg)		
SR1	50	50	-	-	292	8	400		
SR2	50	100	-	-	242	8	400		
SR3	50	150	-	-	192	8	400		
SR4	50	-	50	-	292	8	400		
SR5	50	-	100	-	242	8	400		
SR6	50	-	150	-	192	8	400		
SR7	50	-	-	50	292	8	400		
SR8	50	-	-	100	242	8	400		
SR9	50	-	-	150	192	8	400		

Table 3: Compression of gastro- bilayer floating   tablets.								
Ingredients (mg)	GBF	GBF0	GBF1	GBF2				
Immediate release layer SD equivalent to 10 mg of ezetimibe (ESD1)	40	40	40	40				
CCS	15	15	15	15				
Lactose	44	44	44	44				
Magnesium stearate	1	1	1	1				
Sustained release layer Atenolol	50	50	50	50				
HPMC K100	150	150	150	150				
Sodium bicarbonate	15	20	25	30				
MCC	177	172	167	162				
Magnesium stearate	8	8	8	8				
Total	500	500	500	500				

Each batch contains 50 tablets

#### **Crushing strength**

Crushing strength was determined using Monsanto type hardness tester by selecting randomly six tablets from each batch.

# Friability

Friability was conducted by considering tablets whose weight equivalent to 6.5 g using Roche friabilator apparatus (Model: 40 FT A01). It was rotated for 4 min at 25 rpm where the tablets were allowed to fall from 6 inches height in each turn within the apparatus. These tablets were weighed to calculate percentage friability.

# **Drug content**

Twenty tablets were randomly selected and crushed for the estimation of drug content. Powder weight equivalent to 50 mg was transferred into 50 ml volumetric flask and made to the volume by 0.1 M HCl. The flask was placed in a sonicator till drug completely soluble. The solution was filtered through a filter paper (0.45  $\mu$ m pore size) from this 1 ml was taken and transferred to 25 ml volumetric flask which was made up to the mark by 0.1 M HCl. The absorbance of the solution was measured using UV-Visible spectrophotometer (Agilent, Cary 60) against the blank 0.1 M HCl at 232 nm.

# In vitro floating ability

The buoyancy capability of floating tablets was visually determined by employing the method described by Rosa *et al.*<sup>21</sup> Buoyancy test was carried out by placing tablet in beaker containing 250 ml of 0.1 M HCl. The time required for floating the tablets to appear on to the surface of a dissolution medium (floating lag time, FLT)

and the time during which the dosage form constantly float (total float time, TFT) were measured.

#### **Dissolution studies of gastro-bilayer floating tablets**

Dissolution study was carried out in triplicate for the developed gastro-bilayer floating tabletsusing USP II dissolution test apparatus (Electrolab, Mumbai, Model no. TDT-08L). In dissolution experiments, paddle rotational speed of 50 rpm with 900 ml of 0.1 M HCl dissolution medium was maintained at 37  $\pm$  0.5°C throughout the study. At predetermined time intervals 5 ml of samples were withdrawn for a period of 12 h and filtered through a membrane filter (0.45 µm, Millipore). The volume was replaced with equal volume of fresh dissolution medium maintained at 37  $\pm$  0.5°C after each sampling. The samples withdrawn were analyzed by using a UV-Visible spectrophotometer at 232.4 nm.

### Release kinetics<sup>22-24</sup>

*In vitro* drug release data was subjected to mathematical models like zero order, first order, Higuchi, Hixon Crowell, Korsmeyer and Peppas in order to investigate the release pattern of optimized batch.

### **RESULTS AND DISCUSSION**

Ezetimibe is practically insoluble in water (BCS class II). It has insufficient dissolution rate which is the limiting factor in the oral bioavailability. It has slow absorption leading to inadequate, variable bioavailability and gastrointestinal mucosal toxicity. So, to meet the aim of the study an attempt was made to improve the solubility of ezetimibe by SD technique. Nisha (2016) *et al.*<sup>25</sup> improved the solubility of ezetimibe by preparing SD using PVP K-30 and PEG 6000 by SM. Therefore, for improving the solubility of ezetimibe, SDs techniques were adopted.

# FTIR

The FTIR spectra of ezetimibe, PVP K-30 and SDs were displayed in Fig 1. Ezetimibe was characterized by -OH stretching at 3224.1 cm<sup>-1</sup>, -C = O stretching bands of lactam ring at 1712.7 cm<sup>-1</sup> and -C=C stretching band at 1217.0cm<sup>-1</sup>, -C-O stretching band at 1403.3 cm<sup>-1</sup> (Figure 1A). Nearly similar results were reported by Yasser *et al.*<sup>26</sup> The FTIR spectrum of SD showed similar characteristic absorption peak of ezetimibe and PVP K-30. This indicates no new peaks or major shifting of the peaks denotes that the selected the drug and polymer/excipient are chemically compatible to each other. The physical interaction was necessary for the enhancement of solu-

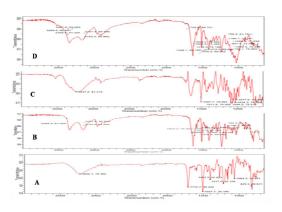


Figure 1: FTIR absorption spectra of A) ezetimibe, B) atenolol, C) ezetimibe SD (ESD1) and D) optimized bilayer tablet.

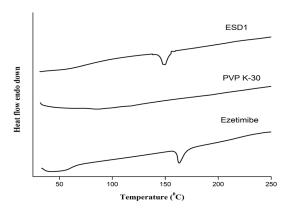


Figure 2: DSC thermograms of ezetimibe, PVP K-30 and ezetimibe SD (ESD1).

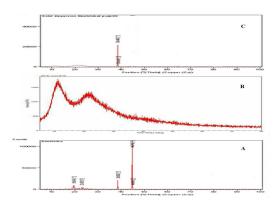


Figure 3: XRD pattern of A) ezetimibe, B) PVP K-30 and C) ezetimibe SD (ESD1).

bility of ezetimibe drug. The intensities of these peaks were reduced in SD mixture and slight shift in the peak values (1709.27 cm<sup>-1</sup>). The physical interaction was identified with PVP K-30 which might be helpful for the enhancement of the solubility of ezetimibe. The FTIR of the bilayer tablet showed no changes in the characteristics of the drugs. This indicates the compatibility between the drugs and excipients.

#### DSC

The DSC thermogram of pure drug ezetimibe showed endothermic peak at 162.66 °C (Figure 2) corresponds to its melting point. This indicates the drug was crystalline in nature. The broadness and shifting of the melting peaks of the drug ezetimibe in SD and decrease in heat of enthalpy indicates the drug was converted into the amorphous state (Figure 2). It confirms that ezetimibe was completely dispersed in the carrier. This was supported by the XRD study. Amorphous states have more surface area and interaction with the solvent. This might be helpful for the enhancement of solubility and dissolution rate of the ezetimibe.

# XRD

The XRD diffractogram patterns of pure drug ezetimibe, PVP K-30 and SD were shown in Figure 3. XRD patterns of pure drug ezetimibe showed sharp and strong intense peaks at 20 equivalent to 19.54°, 23.02°, 38.50°, 44.34° and 44.74° (Figure 3A). This indicates the strong crystalline nature of ezetimibe. PVP K-30 did not show any characteristics intense peaks at 20, indicates the selected polymer was amorphous in nature (Figure 3B). The intensity of peaks at  $2\theta$  equivalent to 38.50° was reduced indicating the conversion to amorphous state. The disappearance of the peaks at  $2\theta$ equivalent to 19.54°, 23.02°, 44.34° and 44.74° (Figure 3C) suggested the transformation to amorphous state. The peaks at 20 equivalent to 19.54°, 23.02° becomes broad. This indicates crystalline nature of ezetimibe was converted to amorphous form. This was supported by DSC study and dissolution study. Amorphous product might be helpful to enhance the solubility and dissolution rate of ezetimibe.

# In vitro dissolution studies

The dissolution profiles of pure drug ezetimibe and SDs were shown in Figure 4. The dissolution value of ezetimibe within 60 min showed an incomplete drug release (21.65  $\pm$  0.48 %). The dissolution profile of SDs showed higher dissolution rate than that of pure drug. Within 5 min of dissolution SD (ESD3) showed approximately 3.39 fold enhanced drug release. There was 59.11 % drug release within 30 min of dissolution. This was approximately 4.04 fold enhancement of drug release than pure drug. It showed 81.64 % drug release within 60 min of dissolution. The similarity factor (f<sub>2</sub>) was found to be 16.38. This indicated the release profile of the SD was different from the pure drug. This was significant (p < 0.05). The enhancement of dissolution of ezetimibe in SDs may be due to the

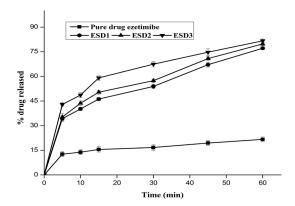


Figure 4: In vitro release profiles for solubility enhancement of ezetimibe.

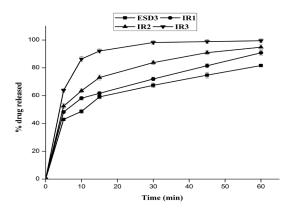


Figure 5: In vitro release profiles of immediate release tablets.

amorphous nature of the drug. Higher free energy of amorphous ezetimibe might have helped better interaction with solvent, leading to enhancement in dissolution. It was supported by XRD and DSC. SD equivalent to 10 mg of ezetimibe (ESD3) was taken and formulated into immediate release tablets by varying the concentrations of superdisintegrants, croscarmellose sodium (5, 10 and 15 % w/w). The powder blends were compressed in direct compression method.

Dissolution profiles of the prepared formulations were depicted in Figure 5. The formulations IR1 and IR2 showed 90.78 % and 94.82 % drug release within 60 min. IR3 formulation containing 15 % w/w of croscarmellose sodium showed 99.49 % within 60 min. It showed 92.08 % of drug release within 15 min (Figure 5). This show nearly 5 fold more dissolution rate when compared to a pure drug over a period of 60 min. This may be due to formulation encounters the dissolution medium, croscarmellose sodium in the immediate release layer of bilayer tablet swells by absorbing the liquid medium by wicking<sup>27</sup> thereby liberating ezetimibe with fine dispersion. Therefore, the IR3 formulation containing 15 % w/w of croscarmellose sodium was selected as immediate

release layer in the preparation of gastro-bilayer floating tablets.

Atenolol is a class III drug with high solubility and low permeability with a half-lie of 6-7 h. Therefore, polymer should be incorporated into the formulation to retard the release. HPMC is a hydrophilic polymer and is strong enough to retard the release depending upon its viscosity and concentration.<sup>28</sup> Formulations with three different grades of HPMC (K4, K15 and K100) in three different concentrations (12.5, 25 and 37.5 % w/w) along with other excipients were accurately weighed. The powder blends were subjected for sustained released tablet in direct compression method. The release profiles of the sustained released formulation (SR1-SR9) were illustrated in Figure 6. Hydrophilic nature of HPMC facilitates penetration of dissolution medium into the network structure of polymer chain thereby causing hydration of the polymer and swelling<sup>18</sup> formed gelatinous layer act as boundary for the drug that must be released. The formulations with HPMC K 100 (SR7, SR8, SR9) showed 100 % of drug release within 9-12 h whereas HPMC K 4 (SR1, SR2, SR3) and HPMC K15 (SR4, SR5, SR6) showed 100 % drug release within 5-7 h respectively which showed in Figure 6. This showed that HPMC K4 and HMC K15 polymers unable to maintain their integrity for longer period when compared to HPMC K100. Therefore, HPMC K100 showed better retardant effect when compared to other two polymers (HPMC K4 and HPMC K15) over a period of 12 h. Among three different concentrations of HPMC K100 the formulation with 37.5 % w/w of polymer (SR9) showed good release profile over a period of 12 h. This may be due to increased concentration of polymer may increase the thickness of gel barrier and tortuosity. It follows release profile as per Robinson and Eriksen equation *i.e.*, 30-35 % of drug release within 1 h and 60- 65 % released within 6 h and remaining drug is released after 12 h.29 Therefore, SR9

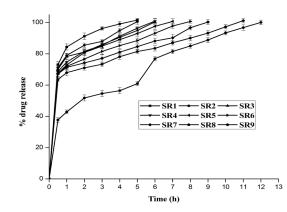


Figure 6: In vitro release profiles of sustained release tablets.

formulation was selected as the optimized formulation. So, SR9 formulation was selected as sustained release layer in the preparation of gastro-bilayer floating tablets. Gastro- bilayer floating tablets were prepared by considering the formulations that showed optimized release profiles from the immediate release layer (IR3) and sustained release layer (SR9) with varying concentrations of sodium bicarbonate by direct compression method and shown in Table 3. The prepared formulations were evaluated for post compression parameters.

# **POST-COMPRESSION PARAMETERS**

General appearance of prepared bilayer tablets was elegant. Differentiation between two layers made by giving blue colour to the immediate release layer and white colour to the sustained layer as shown in Figure 7.

#### Weight variation

Weight variation test is performed to check whether uniform weight is maintained among all the formulation batches. It is affected by the flow properties of the power blend. As weight of the prepared tablet was 500 mg, percentage deviation allowed was 5% as per Indian Pharmacopeia (IP). The prepared tablets were found to be in the range of 496-500 mg (Table 4). So, these results suggested that weight variation was within the range and there was no significant variation in weight between different batches of tablets. This showed uniform die filling during tablet compression.

#### Hardness

Hardness gives an idea regarding how far the tablets resist to capping, abrasion or breakage under conditions of storage, transportation and handling. It affects drug dissolution and release to some extent. The hardness of all the formulations were within the range of 5.5-5.6 kg/cm<sup>2</sup> (Table 4). Therefore, all formulations were within the range. All the batches were found to have good thickness, dissolution and had the ability to withstand the handling abrasion.

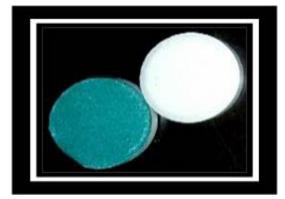


Figure 7: General appearance of prepared gastro- bilayer floating tablets. Immediate release layer (blue) and sustained release layer (white).

# Friability

Friability test gives information regarding percentage loss in weight by tablets due to mechanical stress. Therefore, friability test decides whether the prepared tablets were withstanding the mechanical stress and confirms their suitability for processing. Friability of the prepared tablets were found to be in the range of 0.459-0.612 (Table 4). As friability was below 1 % indicating prepared tablets in each formulation can withstand the mechanical shocks.

#### **Drug content**

Drug content estimation helps in ensuring the consistency of dosage units, in which each unit in a batch should have drug content as per the limits of IP. Assay of gastro-bilayer floating tablets was with the help of UV-Visible Spectrophotometer and the content was estimated. Drug content for all the prepared formulation batches were found in the range of 99.18–100.12 % for ezetimibe and 100.13–100.2 % for atenolol (Table 4) and the values obtained as per IP.

#### In vitro floating ability

Ideally, least possible FLT and continuous flotation of the dosage form in the upper GIT by preventing the dosage form from escaping to the lower GIT is necessary

Table 4: Post-compression parameters of the prepared gastro- bilayer floating tablets.									
Formulation	Weight	Hardness⁵ (kg/cm²)	<b>Friability</b> <sup>c</sup>	Drug con	FLT	TFT			
code	variation <sup>a</sup> (mg)			Ezetimibe	Atenolol	(min)	(h)		
GBF	497±1.13%	5.6±0.21	0.517	99.18±1.18	100.13±1.23	45	7		
GBF0	499±1.74%	5.6±0.19	0.504	99.95±1.71	99.85±1.72	32	7		
GBF1	496±1.10%	5.6±0.17	0.612	99.85±1.11	99.56±0.93	15	8		
GBF2	500±1.11%	5.5±0.19	0.459	100.12±0.99	100.21±0.95	2	12		

a: Avg  $\pm$  % deviation, n=20; b: mean  $\pm$  SD, n=6; c: n=6.5 g; d: mean  $\pm$  SD, n=20

for providing local drug absorption in the stomach. Therefore, the concentration of sodium bicarbonate was critical factor to arrive at the shortest lag time and to prolong the gastric retention time. This may be due to sodium bicarbonate can decrease the density of the tablets as it evolved CO<sub>2</sub> bubbles on reaction with HCl, resulting in the formation of pores due to entrapment of bubbles in the swollen polymer matrices of HPMC which ultimately helped the dosage form to float up to the surface of the medium. In addition, by using HPMC persistent buoyancy was achieved. In vitro floating ability was determined by considering the sodium bicarbonate in different concentrations 6.25 % w/w, 7.5 % w/w. Stages of floating of optimized formulation was shown in Figure 8. The tablet swelled by absorbing the solution at 30 sec. At 60 sec it further absorbing the solution, swelled and tilted. It was partially floated with partial separation of immediate release layer (80 sec). At 121 sec tablet floated and the immediate release layer was completely separated from the tablet. The formulations (GBF and GBF0) showed FLT of 45 min and 32 min and floated up to 7 h. The FLT (15 min) was decreased and floating time was increased as the concentration of the sodium carbonate was increased (6.25 % w/w). Accordingly an increased concentration decreased FLT from 30 min to 2 min. The formulation (GBF2) was considered as the optimized formulation as it has good FLT (2 min) along with TFT (up to 12 h) (Table 4).

#### In vitro dissolution studies

Dissolution profile of the immediate release layer in bilayer floating tablet showed more than 92 % ezetimibe release within 15 min as shown in Figure 9. Atenolol has absorption window in upper GIT whereas poor absorption in lower GIT. This varied absorption results in lowering bioavailability i.e., 50 % with a half-life of 6-7 h. Therefore, was selected to formulate into floating layer. The dissolution profiles were shown in Figure 9. Increasing concentrations of sodium bicarbonate may produce higher level of effervescence that may result in increased rate of pore generation which may results in rapid hydration of the matrices and consequently faster drug release may occur. The formulation GBF, GBF0 and GBF1 showed 72 %, 75 % and 80 % of drug release over a period of 12 h respectively and shown in Figure 9. This may be due to increased rate of pore generation which results in increased rate of drug release. The formulation containing 7.5 % w/w of sodium bicarbonate (GBF2) showed complete drug release over a period of 12 h without losing the integrity of the tablet.

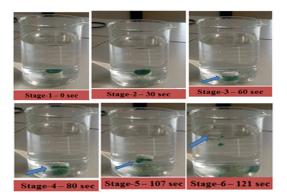


Figure 8: Stages of floating of the prepared gastro-bilayer floating tablets.

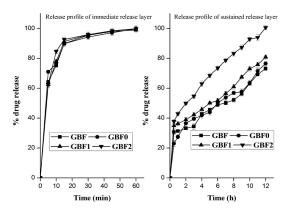


Figure 9: In vitro dissolution profile of gastro-bilayer floating tablets.

Therefore, GBF2 was considered as the optimized formulation.

#### Kinetics of drug release

Drug release kinetics were studied by using zero order, first order, Higuchi, Hixon Crowell, Koresmeyer and Peppas model to optimized formulation and shown in Table 5. Optimized formulation showed bi-phasic release pattern *i.e.*, burst effect followed by sustained release. Burst effect may be due to highly hydrophilic nature of drug, polymer and method of preparation *i.e.*, direct compression method that allows the drug to be on the surface of the tablet without entering the matrix. Therefore, when the tablet comes in contact with the dissolution medium, the drug present on the surface of the tablet entered into medium. Along with this free drug after 1h chain relaxation also occurs for the polymers present at the surface releasing some amount of drug into the dissolution medium (42.86%). So, burst effect is additive effect of the free drug present at the surface and the initial release of drug from the instantly swelled gel barrier. The result further can be supported by the fact that presence of sodium

Table 5: Release kinetics of optimized formulation.										
Formulation	Zero order		First order		Higuchi		Korsmeyer and Peppas		Hixon Crowell	
	<b>r</b> <sup>2</sup>	K <sub>0</sub>	<b>r</b> <sup>2</sup>	K <sub>1</sub>	<b>r</b> <sup>2</sup>	К <sub>н</sub>	<b>r</b> <sup>2</sup>	N	<b>r</b> <sup>2</sup>	К
GBF2	0.870	6.340	0.859	0.077	0.974	25.61	0.969	0.355	0.761	-0.011

bicarbonate also leads to formation of bubbles facilitating the drug release during initial stages.

When the data was subjected to zero order and first order model higher  $r^2$  value was obtained for zero order compared (0.870) to first order (0.859) is high, suggested that formulation follows zero order. This may be due to fact that depleted zone acts as reservoir matrix. The optimized formulation was studied for Higuchi and Hixon Crowell models where Higuchi model  $r^2$  value (0.974) was found to be higher than Hixon Crowell value (0.761), which reveals that the release may follow diffusion mechanism. This may be due to hydrophilic nature of drug incorporated in a semi-solid matrix or sodium bicarbonate that forms gas bubbles on reaction with dissolution medium which creates porous system that allows the drug diffuses through the tortuous pathway created by porous system. Korsmeyer- Peppas model can be applied to any drug delivery system whose release mechanism was not well known or when more than one type of release is involved. Release exponent (n) value of optimized formulation was found to be 0.355 that indicates Fickian diffusion. HPMC being a swellable polymer when it comes in contact with dissolution medium it forms a gel (depletion layer model). Over a period of time, the release is being controlled by this gel layer which acts as a barrier for the drug release. Drug should pass through this barrier gel layer before entering into the dissolution medium. As the process of dissolution continuous, thickness of this barrier gel layer increases so, continuously drug is released and becomes more controlled.

# CONCLUSION

Better hypertension and dyslipidaemia management needs concomitant drug treatment. Ezetimibe complete release was achieved through the SD technique using PVP K-30 (1:3 % w/w) and addition of superdisintegrants croscarmellose sodium. The total floating time of the optimized tablet was 12 h with 9 min of floating lag time. Hydroxypropyl methylcellulose K100 (37.5 % w/w) was optimized as release retardant polymer. Atenolol release was sustained through diffusion mechanism over 12 h. It can be concluded that biphasic drug release pattern was successfully achieved through the formulation of gastro-floating bilayer tablets in this study, allowing strengthened combination therapy for hypertension and dyslipidemia.

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#### **CONFLICT OF INTEREST**

The authors declare no conflict of interest.

# ABBREVIATIONS

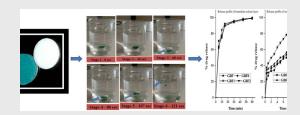
**APIs:** Active pharmaceutical ingredients; **BCS:** Biopharmaceutical classification system; **CCS:** Croscarmellose sodium; **DSC:** Differential scanning calorimetry; **FLT:** Floating lag time; **FTIR:** Fourier transform infrared spectroscopy; **GDDS:** Gastroretentive drug delivery system; **GIT:** Gastro intestinal tract; **HPMC:** Hydroxyl propyl methyl cellulose; **IR:** Immediate release; **MCC:** Microcrystalline cellulose; **NHANES:** National health and nutritional examination survey; **PEGs:** Polyethylene glycols; **PVP:** Polyvinyl pyrrolidones; **SDs:** Solid dispersions; **SM:** Solvent evaporation method; **SR:** Sustained release; **TFT:** Total float time; **XRD:** X-ray diffraction analysis.

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**PICTORIAL ABSTRACT** 

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#### SUMMARY

Hypertension and dyslipidaemia desires combination therapy. Atenolol (short half-life and 50 % absorption) and ezetimibe (BCS class II) were selected as models drug for better therapeutic benefits. Gastro-bilayer floating matrix tablet was formulated (direct compression) by incorporating ezetimibe as immediate release laver after enhancement of solubility (solid dispersion technique-solvent evaporation method) and atenolol as sustained release layer after optimization of its release up to 12 h. Atenolol was released over 12 h through diffusion mechanism where as ezetimibe displayed more than 95 % within 30 min. TFT and FLT showed 12 h and 9 min respectively of the optimized formulation (HPMC K 100-37.5 % w/w and ). This study achieved biphasic drug released and also strengthened the combination therapy.



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