# Quantitative Analysis of Pantoprazole Sodium Sesquihydrate in Bulk and Solid Dosage Form via UV-Spectrophotometric Method

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# ABSTRACT

Pantoprazole sodium sesquihydrate, is a proton pump inhibitor, was analyzed by using UV spectrophotometry. The quantification of pantoprazole sodium sesquihydrate in distilled water was performed in the wavelength range of 290 nm at 20  $\mu$ gmL<sup>-1</sup>. The linearity range is 5-35  $\mu$ gmL<sup>-1</sup> by using UV spectrophotometry. The developed method was applied directly and easily to the analysis of the pantoprazole sodium sesquihydrate in bulk and pharmaceutical tablet preparations. The developed method was calculated in terms of percentage recovery (99.20-101.21%) and %RSD values less than 1% were found in precision. The LOD and LOQ were 0.989 and 1.954  $\mu$ gmL<sup>-1</sup>, respectively. Because of simplicity, accuracy and cost-effectiveness, this validated method is helpful for a daily laboratory analysis of pantoprazole sodium sesquihydrate.

Key words: Pantoprazole sodium sesquihydrate, Proton pump inhibitor, UV-Spectrophotometer, Method development, ICH guidelines.

# INTRODUCTION

Chemically, Pantopraole sodium sesquihydrate is known as sodium 5- (difluoromethoxy) - 2- [3,4 - dimethoxy - 2 - pyridyl) methylsulfinyl] - 1H benzimidazole sesquihydrate.<sup>1</sup> Pantopraole sodium sesquihydrate is a proton-pump inhibitors that block stomach acid by inhibiting the H<sup>+</sup>/K<sup>+</sup>-ATPase enzyme of the secretory caniculus parietal cell thereby curing the acid related problems. As it is unstable at low pH, therefore, available as an enteric coated tablets.<sup>2-4</sup> Pantoprazole sodium sesquihydrate acts as a key drug of triple therapy for Helicobacter Pylori's eradication or individually for the management of esophagus inflammation and maintenance of gastro-esophagus reflux disease, it shows similar pharmacology like other proton pump inhibitors, however shows greater consequence than Histamine H<sub>2</sub> antagonists.<sup>4-8</sup>

Few methods have been reported for the analysis of this drug in literature. Keyur and co-worker also reported the same work for the simultaneous estimation of cinitapride and pantoprazole in pharmaceutical dosage form.9 In 2013 Sourav et al. reported work on the formulation and evaluation of enteric coated tablets of pantoprazole.<sup>10</sup> Another work published on for simultaneous estimation of pantoprazole and levosulpiride using UV spectrophotometric in capsule dosage form.11 Ognjenka with co-worker validated HPLC method for pantoprazole pellets.<sup>12</sup> Utsav et al. reported the dual wavelength method for simultaneous estimation of ondansetron and pantoprazole in combined tablet dosage form.<sup>13</sup>

The published literature survey reveals that there are several analytical methods for the estimation of pantoprazole sodium sesquiSubmission Date: 20-07-2019; Revision Date: 05-11-2019; Accepted Date: 23-12-2019

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hydrate in a combined solid dosage form. But to the best of our knowledge, there is no method for the detection of Pantopraole sodium sesquihydrate in bulk using distilled water as solvent. Therefore, the author have made an attempt to develop modern, rapid, precise and reproducible single analytical method for the detection of pantoprazole sodium sesquihydrate in bulk for first time according to the guidelines stated by USP-39 and International Conference on Harmonization (ICH).<sup>14,15</sup>

## Experimental

#### Instrumentation

UV-Spectrophotometer Model: UV-1800 240 V, Shimadzu, Japan, Electronic Balance: ATL 3000G/0.001G, Dissolution apparatus Model: GDT-6L Serial No: 14-8-95, Galvano Scientific, Pakistan. Electrical Thermostatic Water Bath 220 V 50 Hz, Power (W) 1000±10%, Temperature ranges 37–100°C, Daihan Scientific, China.

## **Reagent and Chemical**

Pantoprazole sodium sesquihydrate kind presented by Bosch Pharmaceuticals (Pvt.) Ltd, while freshly prepared distilled water utilized in the laboratory throughout the experiment. Hydrochloric acid 37% by RCL Lab scan Limited and sodium hydroxide purchased by Daejung Chemicals and Metals Co. Ltd. Zopent tablet (20 mg) of Hilton Pharma (Pvt.) Ltd, Neege Tablet 20 mg of Sami Pharmaceuticals (Pvt.) Ltd., Panizox Tablet 20 mg of S.J. and G. Fazul Ellahie (Pvt.) Ltd. and Zentro Tablet 20 mg by Bosch Pharmaceuticals (Pvt.) Ltd was purchased from a local pharmacy.

### **Preparation of Solutions**

**Stock solution:** 0.02 g of pantoprazole sodium sesquihydrate was weighed accurately, transferred to a volumetric flask (100 mL) and dissolved in 20 mL distilled water. The volume of the flask was made up to the 100 mL and was shaken well. From stock standard solution, 5 mL of pantoprazole sodium was transferred to a volumetric flask (50 mL) and the volume of the flask was made to mark 50 mL with distilled water and was shaken well.

**Preparation of sample solution:** Transfer about 0.02 g of pantoprazole sodium sesquihydrate accurately weighed to a volumetric flask of 100 mL, dissolved in 20 mL distilled water and then diluted with diluents to volume and mixed. After even mixing, the final 20 µgmL<sup>-1</sup> concentrated solutions were prepared by pipetting out 5 mL aliquot of this solution in another 50 mL volumetric flask and diluted it with diluent to calibration mark. Shake the contents to get homogenous dilution.

## MATERIALS AND METHODS

## Analytical method development using UV-Spectrophotometer

Different aliquots were scanned via UV-Spectrophotometer to determinate the wavelength having maximum absorbance.

# Medium suitability of pantoprazole sodium sesquihydrate with distill water

The sample preparation was left up to a specified period of about 24 hrs and stability of pantoprazole sodium sesquihydrate with the distilled water was assessed by comparing solution preparations at different time intervals to that of initial. The solutions were scanned at different time intervals like 2, 4, 8 and 24 hrs and spectra of respective peaks were recorded.<sup>14,15</sup>

## Analytical method validation of proposed method

The validation of proposed method was based on ICH guidelines and USP-39.<sup>14,15</sup>

## Linearity

The calibration curve was plotted in the range of 5 to  $35 \ \mu gm L^{-1}$  to determine the correlation coefficient and y-intercept. The sample solution was serially diluted from 5 to  $35 \ \mu gm L^{-1}$  in distilled water and scanned at 290 nm wavelength to get the absorbance of respective dilution.<sup>16</sup>

#### Precision

Intraday precision was determined by estimating the response of six samples of 20  $\mu$ gmL<sup>-1</sup> on the same day.<sup>14-16</sup>

## Accuracy

The accuracy of the method was determined by calculating the percentage recoveries. So, these 10, 15, 20, 25 and 30  $\mu$ gmL<sup>-1</sup> dilutions were analyzed thrice to assess the percentage recoveries.<sup>14,15</sup>

## Ruggedness

For the determination of intermediate precision 20 µgmL<sup>-1</sup> dilution was analyzed on three consecutive days by three different analysts without changing the parameters and instrument. The results were reported in terms of relative standard deviation %RSD.<sup>14</sup>

#### Robustness

Robustness of the stated method was evaluated by deliberate variation in the solvent and temperature. Deliberately changed tap water instant of distill water and temperature 25°C±5. Scanned the standard solu-

tion following the both samples to find out the robustness of the method. <sup>14</sup>

## Limit of Detection and Limit of Quantification

By using below equation, the LOD of our developed method of pantoprazole sodium sesquihydrate in bulk was determined. Limit of detection [LOD] =3.3  $\sigma$ /S. Limit of quantification was determined by using this equation. Limit of quantification [LOQ] =10  $\sigma$ /S. Specificity of the proposed method was determined by analyzing the standard dilution and the placebo.<sup>16</sup>

# Assay of marketed available pantoprazole sodium tablets

20 tablets from each brands were accurately weighed and grinded them to a fine powder. Transferred powder equivalent to 100mg of Pantopraole sodium sesquihydrate into 50mL volumetric flask. Add 20 mL of distill water and stir to dissolve the contents. Make up the volume to the mark with diluents and mixed well. Then all the dilutions were scanned at 290nm by using UV-Spectrophotometer. Then, absorbance of all the four test solutions was recorded.<sup>17</sup>

## **RESULTS AND DISCUSSION**

### Analytical Method Development Using UV-Spectrophotometer

Solution with 20  $\mu$ gmL<sup>-1</sup> concentration was considered as standard dilution of pantoprazole sodium sesquihydrate and wavelength of 290 nm was selected as quantitative wavelength (Figure 1).

# Medium suitability of pantoprazole sodium sesquihydrate with distill water

It was evaluated that the pantoprazole sodium sesquihydrate with water showed satisfactory behavior during the 24 hr (Table 1) (Figure 2).



Figure 1: Analytical method development using UV-spectrophotometer scheme.

# Analytical method validation of developed method using UV-Spectrophotometer

### Linearity

The linearity of developed method was evaluated at seven different concentrations from 5 to 35 µgmL<sup>-1</sup>. Correlation coefficient, intercept and slope value were also determined for statistical investigation. The observed absorbance of each test sample was plotted against the corresponding concentration and a controlled linear line was obtained (Figure 3) with linear regression equation 0.03 and correlation coefficient 0.99 in the concentration from 5 to 35 µgmL<sup>-1</sup> range. So, it is concluded that the proposed method is linear in the specified range of concentrations.<sup>18</sup>

### Accuracy

Accuracy was determined by analyzing known amounts of analyses and it was calculated as the % recovery of the analytes. Each sample was analyzed three times from concentrations 50, 75, 100, 125 and 150%. Table 2 results revealed that the developed analytical method is highly accurate for the analysis of pantoprazole sodium sesquihydrate in bulk.

### Precision

Six readings of 20 µgmL<sup>-1</sup> were taken as to calculate the intra-day precision. The % recoveries were from 98.81 to 100.52%. On behalf of %RSD range which is less than 1%, it was concluded that the developed method had good precision (Table 3 and Figure 4).

## LOD and LOQ

The distinctive study verifies the no interference of degradation or impurity and solvent in absorbance of Pantoprazole sodium sesquihydrate (Figure 5). As per ICH guidelines, LOD and LOQ were found to 0.989 and 1.954 µgmL<sup>-1</sup>, respectively.



Figure 2: Medium Suitability of Pantoprazole Sodium with Distilled Water.





Figure 3: Linearity curve of developed method.



Figure 4: Precision of developed method.

#### Ruggedness

The ruggedness of the analytical method was estimated by an inter-day experimental condition. Three days practical was performed by three different analysts under the same conditions with the same instrument. The resulting scheme in Table 4 presented a good ruggedness of the method.

#### Robustness

The robustness of the developed method in this experimental work was evaluated by changing the solvent from distilling water to tap water and temperature 25°C  $\pm$  5. The results represented that no critical change was recorded as mention in Table 5.

# Assay of pantoprazole sodium sesquihydrate tablets

According to USP-39, the assay limit of pantoprazole sodium sesquihydrate enteric coated tablet is NLT 90% and NMT 110.0%.<sup>16</sup> Table 5 shows the assay results of Zopent, Neege, Panizox and Zentro. Neege and Zentro have 99.08% and 100.52% close to each other whereas Zopent and Neege showed results above 100%



Figure 5: Specificity of Developed Method.



Figure 6: Assay of Pantoprazole Sodium Tablets.

(Table 6, Figure 6). All the % availabilities were in limit and all brands pass the assay results.

#### CONCLUSION

A simple, selective, rapid, accurate and precise analytical method by using UV-spectrophotometer for estimation of pantoprazole sodium sesquihydrate in bulk was developed and validated according to USP-39 and ICH guidelines. Commonly and easily available solvent and

Table 1: Medium suitability of pantoprazole sodiumwith distilled water							
s	Time (hrs)	Abs	%Pecoverv				
No.		Sample	Standard	/intecovery			
1	2	0.758	0.767	99.82			
2	4	0.766	0.767	99.86			
3	6	0.766	0.767	99.86			
4	8	0.76	0.767	99.73			
5	24	0.768	0.767	100.13			

Table 2: Accuracy of Developed Method Using UV-Spectrophotometer						
C No.	Concentration		Ab	osorbance	% Decourant	
5 NO.	µg/mL	%	Sample Standard		%Recovery	
			0.383	0.382	100.26	
			0.384	0.382	100.52	
1	10 ( <i>n</i> =3)	50	0.385	0.382	100.78	
			0.575	0.573	100.34	
			0.580	0.573	101.21	
2	15 ( <i>n</i> =3)	75	0.579	0.573	101.04	
			0.759	0.764	99.34	
			0.759	0.764	99.34	
3	20 ( <i>n</i> =3)	100	0.765	0.764	100.13	
			0.951	0.958	99.26	
			0.959	0.958	100.10	
4	25 ( <i>n</i> =3)	125	0.950	0.958	99.16	
			1.150	1.146	100.34	
			1.149	1.146	100.26	
5	30 ( <i>n</i> =3)	150	1.146	1.146	100.00	

Table 3: Precision of Developed Method Using UV-Spectrophotometer							
S No.	Concentration		Absorbance		~-	~	
	µgmL⁻¹	%	Sample	Standard	%Recovery	%RSD	
1	20	100	0.766	0.762	100.52		
2	20	100	0.763	0.762	100.13		
3	20	100	0.765	0.762	100.39	0.625	
4	20	100	0.753	0.762	98.81	0.625	
5	20	100	0.761	0.762	99.86		
6	20	100	0.759	0.762	99.60		

Table 4: Ruggedness of Developed Method Using UV-Spectrophotometer							
S No.	Analyst	Day	Absorbance				
			Sample	Standard	%Recovery	/0 K3D	
1		Day-1	0.761	0.760	100.13	0.370	
2	A		0.761		100.13		
3			0.759		99.86		
1		Day 2	0.769	0.769	100.00		
2	В		0.768		99.86		
3			0.767		99.73		
1		Day 3	0.765	0.766	99.86		
2	С		0.762		99.47		
3			0.758		98.95		

Table 5: Robustness of Developed Method								
S	Concentration		Solvent	Temperature (°C)	Absorbance		% Pecoverv	
No.	µg/mL	%	Solvent		Sample	Standard	70 Necovery	
1 20		100	Tap Water	20°C	0.763	_	99.35	
	20			25 °C	0.768		100.00	
			30°C	0.770	0.768	100.26		
2 20		100	D.I Water	20°C	0.765		99.61	
	20			25 °C	0.769		100.13	
				30°C	0.772		100.52	

Table 6: Assay of Pantoprazole Sodium Tablets								
S No.	Brand	Absorbance	Limit	Result				
1.	Zopent Tablet	0.806	*NLT 90% and NMT	105.35				
2.	Neege Tablet	0.758	110%	99.08				
3	Panizox Tablet	0.812		106.14				
4.	Zentro Tablet	0.769		100.52				

• Not less than (NLT) and Not more than (NMT)

technique was adopted for the experimental work. So, we can say that the developed analytical method is also cost-effective. On the basis of above results and discussion, the developed method for determination of pantoprazole sodium sesquihydrate in bulk can be used in routine analysis. The pantoprazole sodium sesquihydrate solution is stable up to 24 hrs under normal environmental condition. Moreover, the developed method is also successfully applied for the determination of pantoprazole sodium sesquihydrate in solid dosage form. The marketed brands passed the pharmacopocial test.

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### **CONFLICT OF INTEREST**

The authors declare no conflict of interest.

## **ABBREVIATIONS**

**LOD:** Limit of detection; **LOQ:** Limit of quantification; **mL:** Milliliter; **UV**: Ultra violet.

### REFERENCES

 Suslu I, Altinoz S, Yildiz E. Determination of Pantoprazole in Tablet Dosage Forms by Two Different Spectrophotometric Methods. Fabad J Pharm Sci. 2003;28(2):85-92.

- El-Sharif ZA, Mohamed AO, El-Bardicy MG. Reversed-Phase High Performance Liquid Chromatographic Method for the Determination of Lansoprazole, Omeprazole and Pantoprazole Sodium Sesquihydrate in Presence of Their Acid-Induced Degradation Products. Chem Pharm Bull. 2006;54(6):814-8.
- Basavaiah K, Kumar A, Kalsang TUR. Spectrophotometric Determination of Pantoprazole Sodium in Pharmaceuticals Using N-Bromosuccinimide, Methyl Orange and Indigo Carmine as Reagents. Iran J Chem Chem Eng. 2009;28(1):31-6.
- Cheer SM, Prakash A, Faulds D, Lamb HM. Pantoprazole: An Update of Its Pharmacological Properties and Therapeutic Use in the Management of Acid-Related Disorders. Drugs. 2010;63(1):101-33.
- Sandesh PK, Prashant JP. Bioequivalence Study of Pantoprazole Sodium-HPBCD and Conventional Pantoprazole Sodium Enteric-Coated Tablet Formulations. ISRN Pharmacology. 2013;347457.
- Sourav T, Mahantesh A, Sabitha JS, Rinku M, Prasanth VV. Formulation and Evaluation of Enteric Coated Tablets of Pantoprazole. International Journal of Pharmaceutical and Chemical Sciences. 2013;2(3):1454-61.
- Sahdev C, Vishal B, Hiren K. Development and Validation of UV Spectrophotometric Method for Simultaneous Estimation of Pantoprazole and Levosulpiride in Pharmaceutical Dosage Form. American Journal of Pharmtech Research. 2013;3(1).
- Ognjenkarahic EV, Indira M, Jasminahadžiabdic AE. Development and Validation of HPLC Method for Determination of Pantoprazolin Pantoprazole Pellets. International Journal of Pharmacy Teaching and Practices. 2013;4(4):793-6.
- Keyur B, Dharti N, Charmee B. Analytical method development and validation for simultaneous estimation of cinitapride and pantoprazole in pharmaceutical dosage form. Der Pharma Chemica. 2014;6(2):252-7.
- Tribedi S, Tribedi M, Sabitha JS. Formulation and Evaluation of Enteric Coated Tablets of Pantoprazole. International Journal of Pharmaceutical and Chemical Sciences. 2013;2(3):1454-61.
- Shah DA, Patel A, Baldania SL. Simultaneous Estimation of Pantoprazole Sodium and Levosulpiride in Capsule Dosage Form by Simultaneous Equation Spectrophotometric Method. Hindawi Publishing Corporation. 2013;4:1-4.
- Rahic O, Vranic E, Mujein I. Development and Validation of HPLC Method for Determination of Pantoprazole in Pantoprazole Pellets. International Journal. 2013;4(4):793-6.

- Utsav DP, Falgun AM, Dimal AS, Usman GK, Kashyap KB. Development and Validation of Dual Wavelength Method for Simultaneous Estimation of Ondansetron and Pantoprazole in Combined Tablet Dosage Form. International Journal of Pharmacy and Integrated Life Sciences. 2013;2(I5);101-13.
- 14. USP 39 Transfer of Analytical Procedures/ General Information. 2005;1641-9.
- International Conference on Harmonization of Technical Requirements For Registration of Pharmaceuticals For Human Use Validation of Analytical Procedures: Text and Methodology Q2(R1) Current Step 4 Version Parent



Guideline. (Complementary Guideline on Methodology Dated 6 November 1996 Incorporated In November 2005). 1994.

- Somia G, Kashifa K, Nusrat M. New validated method for analysis of salymarin in polyherbal formulation (aqueous extract, oral liquid and solid dosage form). Chemistry International. 2015;1(3):103-6.
- 17. USP 39–NF34 Page 5269 Pharmacopeial Forum: USP Monographs: Pantoprazole Sodium Delayed-Release Tablets. 2005;33(6):1197.
- Somia G, Sania B. U.V Visible Spectrophotometric Determination, Validation and *in vitro* Interaction of Rabeprazole with Chlorazepate. Saudi J Med Pharm Sci. 2017;3(8):813-7.

#### SUMMARY

Simple, accurate and cost-effective UV spectrophotometric method has been reported for the estimation of pantoprazole sodium sesquihydrate which is potentially useful for a routine analysis for pharmaceutical and clinical laboratories.

#### About Authors



Asma Rafiq has completed her M.Phil., with specialization in the field of Pharmaceutical method development via spectroscopy and stability studies. Currently she is working as an Assistant Manager Regulatory Affairs, Nabi Qasim Industries Pvt. Ltd. She actively participates in research and presents her research work in different National and International conferences.



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