

# Application of the experimental design method to photostability studies of Karvileks tablet

Ivana Savic<sup>1\*</sup>, Valentina Marinkovic<sup>2</sup>, Ivan Savic<sup>1</sup>, Predrag Sibinovic<sup>3</sup> and Nebojsa Cekic<sup>4</sup>

<sup>1</sup>Faculty of Technology, Univerzity of Nis, Bulevar oslobodjenja 124, Leskovac, Serbia

<sup>2</sup>Faculty of Pharmacy, Vojvode Stepe 450, Belgrade, Serbia

<sup>3</sup>Pharmaceutical and Chemical Industry - Zdravlje Actavis, Vlajkova 199, Leskovac, Serbia.

<sup>4</sup>DCP Hemigal, Leskovac, Serbia.

## ABSTRACT

Submitted: 06/07/2011

Revised: 12/09/2011

Accepted: 18/12/2011

This study was examined a photostability of Karvileks (Zdravlje-Actavis, Serbia), packed in the different primary packs. For these examination, the experimental design with 3 independent variables: wavelength of radiation ( $X_1$ ), duration of radiation ( $X_2$ ) and type of packs ( $X_3$ ) was applied. After irradiation of Karvileks, the content of carvedilol and its impurities A, B and C was taken as a response ( $Y_1, Y_2, Y_3, Y_4$ ). The obtained mathematical equations can be used for determination of carvedilol and its impurities content in any case. Tablets were packed in an opaque plastic container, as well as in red blister. The content of carvedilol was determined using a stability-indicating RP-HPLC method. The obtained models were:  $Y_1 = 93.11 + 1.74 X_1 - 2.19 X_2 + 2.01 X_3 + 0.39 X_1 X_2 + 0.39 X_1 X_3$ ;  $Y_2 = 0.22 - 0.02 X_1 + 0.14 X_2 - 0.01 X_3 - 0.01 X_1 X_2 - 0.007 X_2 X_3$ ;  $Y_3 = 0.001 + 0.014 X_1 + 0.014 X_2 - 0.001 X_3 + 0.014 X_1 X_2 - 0.001 X_2 X_3$ ;  $Y_4 = 0.102 + 0.008 X_1 + 0.064 X_2 - 0.007 X_3$ . By using an experimental design, the opaque plastic container was selected as an adequate primary pack for Karvileks. This pack provides an adequate protection of the preparation from the effect of sunlight.

**Keywords:** experimental, design, photostability, Karvileks.

## INTRODUCTION

The main use of experimental design is to enable a simultaneous, systematic and rapid evaluation of each factor effect on the selected responses of the parameters. Also, it can be predicted the dependence of the effects between individual factors, identification of the critical factors, as well as its levels<sup>1</sup>.

The experimental design has found wide application in the formulation of pharmaceutical products<sup>2-9</sup>, optimization of process phases<sup>10</sup>, development of analytical techniques<sup>11</sup>, as well as in studies of drug stability<sup>12-20</sup>.

In this paper, it was investigated product Karvileks (Zdravlje-Actavis, Serbia) from group of beta blockers. Carvedilol, as an active substance of product, is a photosensitive<sup>21</sup>. In literature, it can be found information about chromatographic methods<sup>22-34</sup>, which used for determination of carvedilol concentration in biological fluids (blood, plasma and serum). Many methods are focused on the separation of enantiomers<sup>26-30</sup>. P. Sripalakit and coworkers<sup>35</sup> were developed a carvedilol assay in tablet dosage form using HPLC with fluorescence detection. M. Rizwan et. al. were developed a *stability-indicating* LC method<sup>36</sup> to study the kinetics of carvedilol

degradation in accordance with ICH regulative<sup>37</sup>. Stojanovic et al.<sup>21</sup> were developed and validated RP-HPLC method for simultaneous separation and determination of carvedilol, as well as its potential impurities in pharmaceutical formulation. They also monitored its photochemical stability.

The aim of this paper was to examine a photostability of Karvileks (Zdravlje-Actavis, Serbia), packed in different primary packs. For these examination, it was applied the model of experimental design with 3 independent variables factors: wavelength of radiation ( $X_1$ ), duration of radiation ( $X_2$ ) and type of packs ( $X_3$ ). The content of carvedilol and its impurities A, B and C were taken as a response ( $Y_1, Y_2, Y_3, Y_4$ ), after irradiation of Karvileks.

## MATERIAL AND METHODS

**Samples.** Standard of carvedilol was purchased from NOSCH-a Labs Private Limited, India. Standards of impurities A, B and C were purchased from Merck, Darmstadt, Germany. Karvileks tablets, manufactured by Zdravlje-Actavis, Serbia, is contained a 12.5 mg of carvedilol.

**Reagents.** Acetonitrile HPLC grade was purchased from Merck, Darmstadt, Germany. All chemicals used were of analytical grade and deionized water was HPLC grade.

**Photostability studies.** Analyzed tablets were packed in an opaque plastic blister (Al/PVC) and red blister. They are exposed to light to determine the effects of irradiation on the stability of the drugs. All samples for photostability testing

### \*Address for Correspondence:

Dr Ivana Savic, Department of Pharmaceutics, Faculty of Technology, Bulevar oslobodjenja 124, 16000 Leskovac, Serbia.

E-mail: vana.savic@yahoo.com

were placed in a light cabinet (Suntest CPS/CPS<sup>+</sup>, Atlas Material Testing Technology, Germany) and exposed to different wavelength (254 and 540 nm) and duration of radiation (3.5; 7 and 17 days) at 25 °C. Control samples which were protected from light were also placed in the light cabinet and exposed concurrently. Following removal from the light cabinet, all samples were prepared for analysis as described in the part for preparation of tablets.

**Apparatus.** The method was performed by Agilent 1100-Series HPLC system (Agilent Technologies, USA), consisting of a HP G13141A variable wavelength UV detector and Agilent 1100-Series auto-sampler using a 20 µL sample loop. The system was controlled and data analyses were performed by Agilent HPLC Data Analysis software. The reproducibility was performed by another LC system, consisting of an Agilent 1100-Series binary pump and Agilent 1100-Series DAD detector (Faculty of Technology, Leskovac). The detector was set at 240 nm and the peak areas were integrated automatically using the Agilent HPLC Data Analysis software program. The separation of carvedilol from its impurities was carried out at 20 °C using an octylsilyl silica column (4.6 x 100 mm, 5 µm) Agilent Technologies, USA.

**Chromatographic conditions.** RP-HPLC analysis was performed by isocratic elution with a flow rate of 1 mL/min. A mobile phase was contained a phosphate buffer in water : acetonitrile (65:35 v/v). Procedure for preparation of phosphate buffer is the following: dissolve a 1.77 g potassium dihydrogen phosphate in 650 mL of water. All solvents were filtered through a 0.45 µm millipore filter. Volumes of 20 µL of the solutions and samples were injected into the column. In accordance with the Eur Ph VI<sup>38</sup>, the expected retention time of carvedilol was 4.5 min. Retention time of impurities A, B and C were 0.6 min, 3.5 min and 6.7 min, respectively.

**Packaging.** Two container systems (blister and opaque plastic packs) were used in the stability studies. The first was a blister pack comprising 200 µm PVDC (red and white colour) of grade 40 g/m<sup>2</sup> and of 20 µm aluminium foil. These packs were purchased from Ceduliose (St Maur des Fossés, France). Packs were formed on the Noack DPN 760 blister packers; the plastic film was unwound from the reel and guided through a preheating station on the blister line. The temperature of the pre-heating plates (for the upper and lower plates this is fixed at about 120 °C) was such that the plastic became soft and mouldable and cavities were then formed in a forming station using compressed air. Finally, the aluminium foil was heat-sealed at about 170 °C.

The second tested container was the opaque plastic container. Plastic container content was HDPE (high-density polyethylene). The mass density of HDPE can range from 0.93 to 0.97 g/ml. It is also hard and opaque and can withstand somewhat higher temperatures (120 °C for short periods, 110 °C continuously).

**Preparation of standard solutions.** 12.5 mg of carvedilol and 10 mg of impurities A, B and C were dissolved in 10 mL volumetric flask in mobile phase, with mixing in the ultrasonic bath for 15 min. 1 mL of this solution was transferred into another clean dry flask of 10 mL. 1 mL of this solution contained 0.125 mg of carvedilol and 0.1 mg of impurities A, B and C.

**Sample preparation of product Karvileks.** After radiation, tablets were finely broken and all the powder content was transferred into conical flask of 100 mL. The obtained solution was inserted in the ultrasonic bath and sonificated for 15 min. After that, the obtained solution was filtered through qualitative filter paper and then through a membrane filter of 0.45 µm. 1 mL of this solution was transferred into flask of 10 mL and supplemented by mobile phase to the mark. A non-irradiated sample of tablets was used as a blank sample.

#### Equations for calculating a content of carvedilol and impurities

For determination a content of carvedilol and impurities in the tested samples, it was used a following equation was used:

$$\text{mg Carvedilol/ tablet} = \frac{I_p \times W_{st} \times K}{1st \times 100} \dots\dots\dots (1)$$

$$\% \text{ Impurity A} = 2 \times A_{\text{analyzed solution (which correspond to impurity A)}} \times 0.5 / 2 \times A_{st} \dots\dots\dots (2)$$

$$\% \text{ Impurity C} = A_{\text{analyzed solution which correspond to impurity C}} \times 0.05\% / 10 \times A_{stc} \dots\dots\dots (3)$$

$$\% \text{ other impurity} = A_{isp} \times 0.5 / A_{st} \dots\dots\dots (4)$$

where is:

$I_p$  – peak area of carvedilol in the tested solution

$I_{st}$  – peak area of carvedilol in standard solution

$W_{st}$  – measured amount of standard substance of carvedilol

$A_{st}$  – peak area of standard solution of impurity A

$A_{stc}$  - peak area of standard solution of impurity C

**Experimental design.** For studying the influence of primary packs on photostability of Karvileks, eight experiments were prepared. For this investigation, tablets were packed in a red blister and opaque plastic container. After radiation, the content of carvedilol and its impurities, as response ( $Y_1$ ,  $Y_2$ ,  $Y_3$  and  $Y_4$ ), were determined using a stability-indicating RP-HPLC method.

The data was statistical analyzed using Statistica 8.0 (Stat Soft, Inc., Tulsa, USA). The values of response and independent variables of model experimental design are presented in Table No. 1.

Table 1: The combination of independent variables and the value of dependent variables

Run	X <sub>1</sub> , nm	X <sub>2</sub> , day	X <sub>3</sub>	Y <sub>1</sub> , Content of carvedilol (%)	Y <sub>2</sub> , Content of Impurity A (%)	Y <sub>3</sub> , Content of Impurity B (%)	Y <sub>4</sub> , Content of Impurity C (%)
1	254	3.5	red blister	92.1	0.09	0	0.04
2	540	3.5	red blister	94.6	0.08	0	0.04
3	254	17	red blister	87.4	0.401	0	0.173
4	540	17	red blister	90.3	0.362	0.059	0.183
5	254	3.5	plastic container	95.8	0.08	0	0.03
6	540	3.5	plastic container	98.7	0.07	0	0.04
7	254	17	plastic container	90.2	0.371	0	0.132
8	540	17	plastic container	95.8	0.309	0.049	0.176

## RESULTS AND DISCUSSION

Environmental factors, such as temperature, humidity and light, are the main causes of drug degradation.

The effect of sunlight on the drug can have a negative impact on its quality, safety and pharmacotherapy. For these reasons, all researches in the direction of photodegradation a Karvileks (Zdravlje Actavis, Serbia) are justified.

In the literature can be found information about monitoring photochemical stability of Karvileks by long-term methodes<sup>21</sup>. The product was kept at room temperature, exposed to the effect of sunlight. The percentage of degradation products was measured every 20 days until 100<sup>th</sup> day.

In addition to long-term methods for investigating photostability of carvedilol, Rizwan et al.<sup>36</sup> was applied an accelerated aging test in accordance with ICH regulative. For this investigation, methanolic solutions of carvedilol were prepered, which were exposed to the sunlight (72 h) and UV radiation at a wavelength of 254 nm for 3 h.

Having in the mind the fact that carvedilol is a photo-sensitive substance, the influence of different primary packs, as well as effect of wavelength and duration radiation on tablet stability was tested.

For studying a stability of product, it can be applied several methodologies. In this paper, methodology of experimental design was applied for investigation an influence of wavelength (X<sub>1</sub>) and duration (X<sub>2</sub>) of radiation and type of packs (X<sub>3</sub>) on product stability of Karvileks (Zdravlje-Actavis, Serbia). The advantage of this method is simultaneous, systematic and rapid evaluation of the independent variables effect to the appropriate response (the content of carvedilol and impurities). By using the experimental design, the interaction between variables can be defined. The statistical analysis is the best indicator of the critical independent variables. Also, the limits of variables can be predicted using the experimental design.

The coefficients of proposed model were calculated, which represent the relationship between response and independent variables.

The mathematical model is presented in the form of follows equations:

$$Y_1 = 93.11 + 1.74 X_1 - 2.19 X_2 + 2.01 X_3 + 0.39 X_1 X_2 + 0.39 X_1 X_3 \dots \dots \dots (5)$$

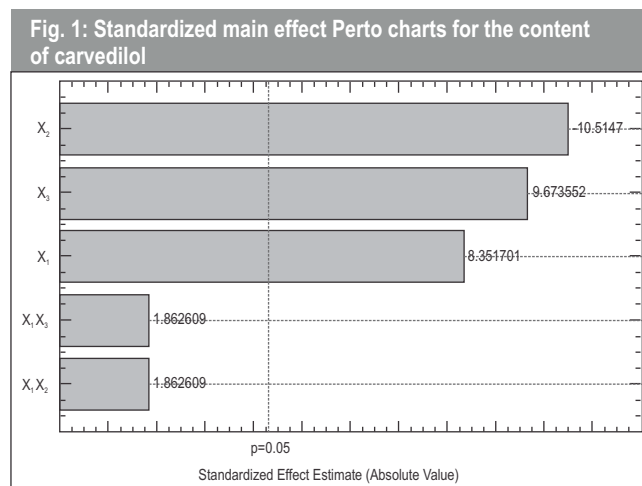
$$Y_2 = 0.22 - 0.02 X_1 + 0.14 X_2 - 0.013 X_3 - 0.01 X_1 X_2 - 0.007 X_2 X_3 \dots \dots \dots (6)$$

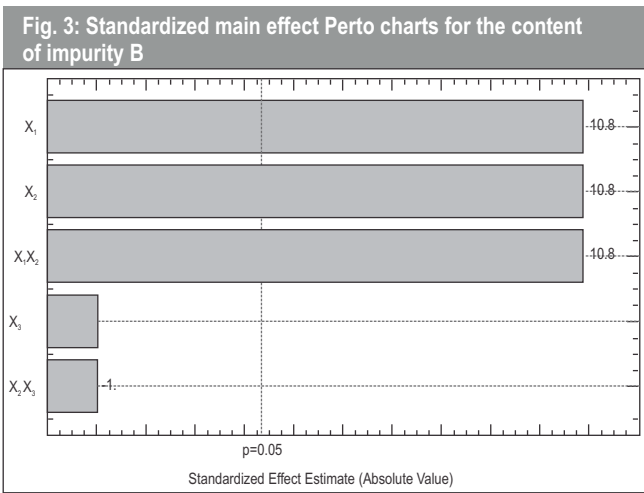
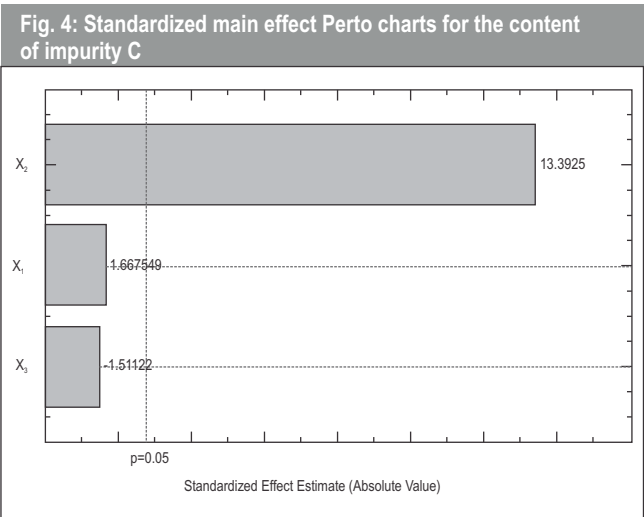
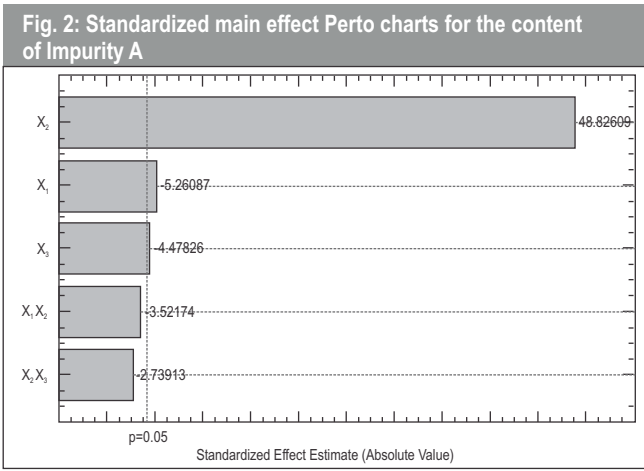
$$Y_3 = 0.001 + 0.014 X_1 + 0.014 X_2 - 0.001 X_3 + 0.014 X_1 X_2 - 0.001 X_2 X_3 \dots \dots \dots (7)$$

$$Y_4 = 0.102 + 0.008 X_1 + 0.064 X_2 - 0.007 X_3 \dots \dots \dots (8)$$

On the basis of the mathematical model and statistical data using ANOVA, the significance of equation terms can be estimated. When the value of p is lower than 0.05, the equation term is the significant. These terms have a significant impact on the content of carvedilol and its impurities in the tested preparation. The equation 5-8 has the values of correlation coefficient of 0.993, 0.999, 0.980, and 0.963, respectively.

The main effects of variables on the content of carvedilol and its impurities (A, B, C) are shown in Figs. 1-4.





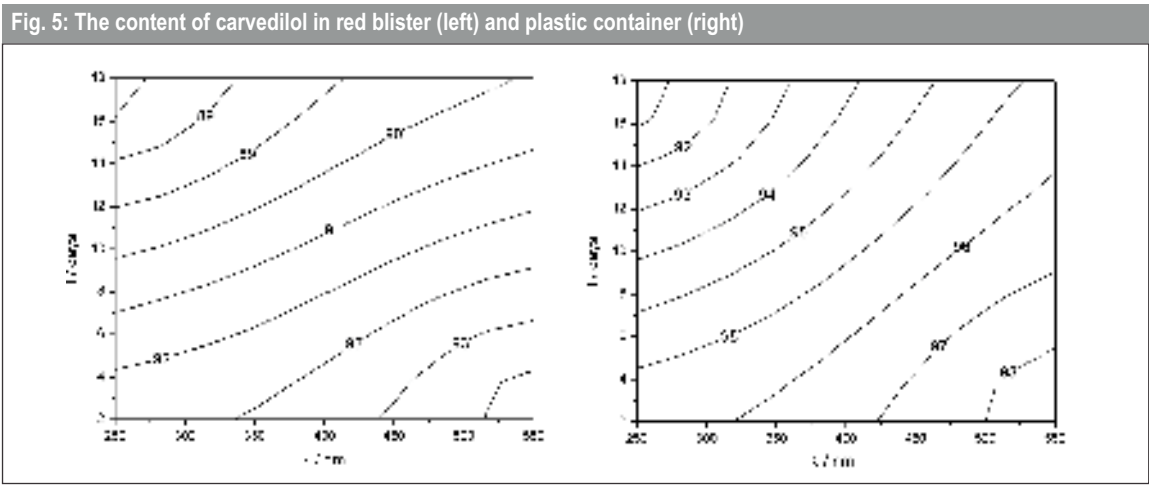
In the case of a change just one independent variable (Fig. 1), it was noted that the duration of radiation ( $X_2$ ) had a dominant influence on the content of carvedilol. Type of packs ( $X_3$ ) had less impact, while wavelength of radiation ( $X_1$ ) had a negligible effect compared with the previously discussed factors. The greatest influence on the change of impurities

content had duration of radiation ( $X_2$ ) and type of radiation ( $X_1$ ).

A relationship between response and independent variables was shown using the contour diagrams (Figs. 5-8). These diagrams are useful for illustrating the influence of factors independent variables on the change of carvedilol content and its impurities in the irradiated samples of Karvileks.

The preparation was relatively photostable in the wavelength range from 400 to 550 nm for both types of primary packs (Fig. 5). Because, the content of carvedilol in irradiated samples of Karvileks was over 90%, accordance with ICH Q1B regulative. The carvedilol content was in the range of 92 to 95% for a red blister, while the content (96.56 to 99%) was slightly higher for a plastic container at the same value of independent variable  $X_1$ .

The content of carvedilol was of 87 to 92% for red blister and slightly higher (92.5 to 95.75%) for opaque plastic container at the change of variable  $X_1$  in the range of 250 to 400 nm.



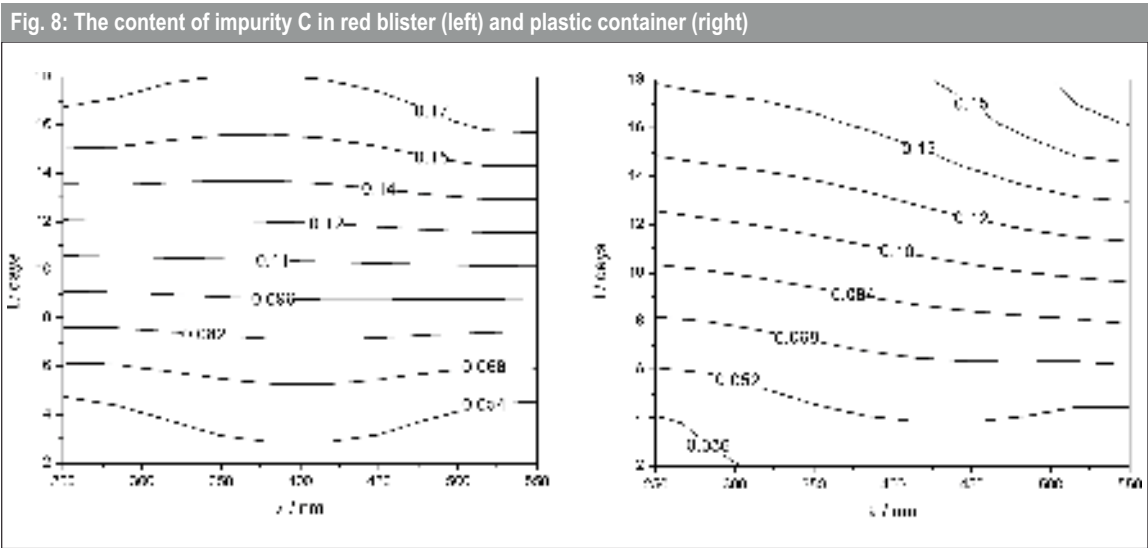
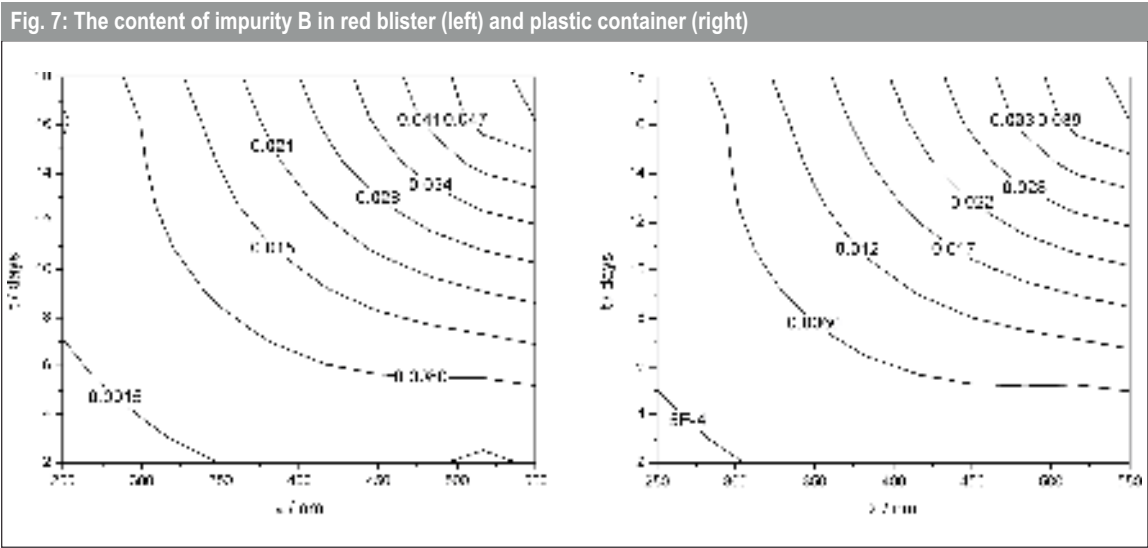
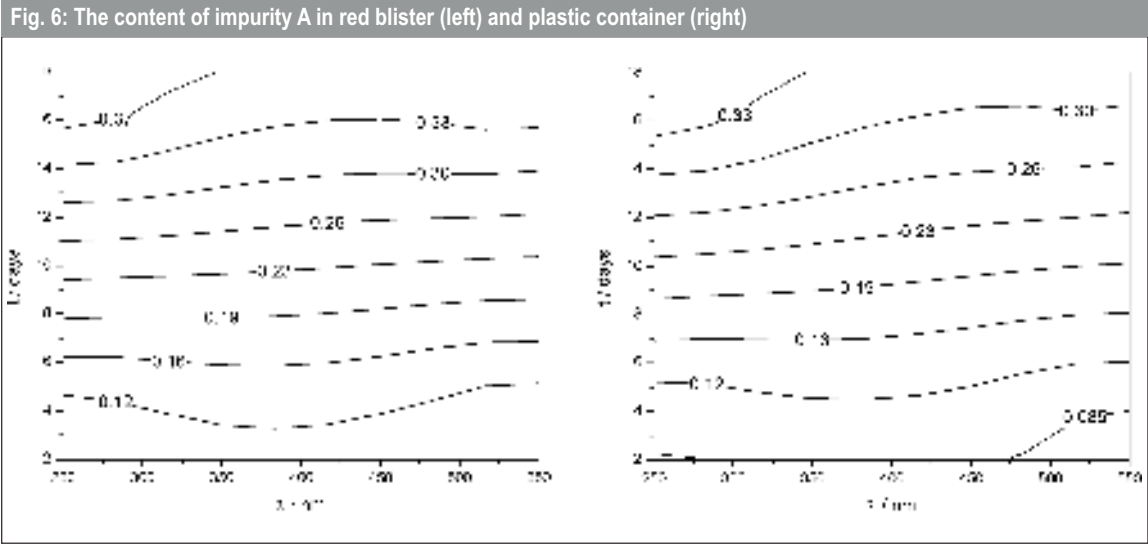


Table 2: The statistical analysis (ANOVA) of response

Variable	df	SS	MS	F value	p value
<b>Y<sub>1</sub> (Content of carvedilol)</b>					
Model	7	97.9	97.9	14.209	0.20
X <sub>1</sub>	1	24.2	24.2	36.524	0.10
X <sub>2</sub>	1	38.3	38.3	57.892	0.08
X <sub>3</sub>	1	32.4	32.4	49.000	0.09
X <sub>1</sub> X <sub>2</sub>	1	1.2	1.2	1.816	0.41
X <sub>1</sub> X <sub>3</sub>	1	1.2	1.2	1.816	0.41
X <sub>2</sub> X <sub>3</sub>	1	0.0	0.0	0.047	0.86
X <sub>1</sub> X <sub>2</sub> X <sub>3</sub>	1	0.7	0.7	1	0.50
Error	1	0.7	0.7	-	-
<b>Y<sub>2</sub> (Content of Impurity A)</b>					
Model	7	0.2	0.2	14.471	0.03
X <sub>1</sub>	1	0	0	27.677	0.12
X <sub>2</sub>	1	0.2	0.2	23.84	0.01
X <sub>3</sub>	1	0	0	20.055	0.14
X <sub>1</sub> X <sub>2</sub>	1	0	0	12.403	0.18
X <sub>1</sub> X <sub>3</sub>	1	0	0	1	0.5
X <sub>2</sub> X <sub>3</sub>	1	0	0	7.503	0.22
X <sub>1</sub> X <sub>2</sub> X <sub>3</sub>	1	0	0	1	0.50
Error	1	0	0	-	-
<b>Y<sub>3</sub> (Content of Impurity B)</b>					
Model	7	0	0	-	0.0001
X <sub>1</sub>	1	0	0	116.64	0.06
X <sub>2</sub>	1	0	0	116.64	0.06
X <sub>3</sub>	1	0	0	1	0.5
X <sub>1</sub> X <sub>2</sub>	1	0	0	116.64	0.06
X <sub>1</sub> X <sub>3</sub>	1	0	0	1	0.5
X <sub>2</sub> X <sub>3</sub>	1	0	0	1	0.5
X <sub>1</sub> X <sub>2</sub> X <sub>3</sub>	1	0	0	1	0.5
Error	1	0	1	-	-
<b>Y<sub>4</sub> (Content of Impurity C)</b>					
Model	7	0	0	0	0.0001
X <sub>1</sub>	1	0	0	7.11	0.23
X <sub>2</sub>	1	0	0	458.67	0.03
X <sub>3</sub>	1	0	0	5.840	0.25
X <sub>1</sub> X <sub>2</sub>	1	0	0	3.361	0.32
X <sub>1</sub> X <sub>3</sub>	1	0	0	3.361	0.32
X <sub>2</sub> X <sub>3</sub>	1	0	0	2.507	0.36
X <sub>1</sub> X <sub>2</sub> X <sub>3</sub>	1	0	0	1	0.50
Error	1	0	0	-	-

The obtained results indicate that Karvileks show a greater stability in VIS field compared with UV.

In addition, the wavelength and duration of radiation had an influence on the photostability of product. It was noted from diagram that the product was relatively photo stable for 7 days of duration of radiation, in the case of a red blister and VIS

radiation (400 to 550 nm), or 13 days for plastic container. Also, it was confirmed that the type of packs had a significant influence on stability of Karvileks. For both wavelength of radiation (254 and 540 nm), the content of carvedilol was higher using a plastic container compared with a red blister.

The obtained contour diagrams, where are presented the content of pharmacopoeia impurities A, B and C (Figs. 6-8), there are zones where the impurities content was less than 0.1%.

With contour diagrams can be noted that duration of radiation and type of packaging had the greatest impact on content change of impurities, which was confirmed using statistical analysis.

The type of packs and duration of radiation have the highest influence on the change of impurities content, what is confirmed using ANOVA test.

The content of impurity A, when the product was packed in a red blister, was less than 0.1% for radiation in the range of 480 to 550 nm and for 3 days of duration of radiation. Using of a plastic container, the content of the same impurity was remained less than 0.1% using radiation in wider range of wavelength (425 to 550 nm) and longer duration of radiation (6 days).

The content of impurity C was less than 0.1% for a red blister, when the radiation wavelength was in the range of 250 to 550 nm for 9h. But the same content of impurity C was achieved after 11 days of radiation for a plastic container.

Unlike to the impurities A and B, the content of impurity B was less than 0.1% in all investigated cases. The obtained low values were indicated on a high photostability of this impurity.

## CONCLUSION

Using of the experimental design, it was found that the dominant influence on the content of carvedilol and its impurities had duration of radiation (X<sub>2</sub>). It was found that the greatest change of the carvedilol content and its impurities was been during simultaneous change of the wavelength and duration of radiation (X<sub>1</sub>X<sub>2</sub>). The most adequate primary packs for Karvileks compared with red blister. On the basis of the content carvedilol and its impurities, it was confirmed that the opaque plastic container is an adequate pack for protection a product from the effect of daylight. In this case the preparation was shown as a relatively photostable.

**Acknowledgments.** The authors are grateful to the colleagues of Pharmacy and Chemical Industry Zdravlje-Actavis, Serbia for the given practical help. This work was supported by the Ministry of Education and Science of the Republic of Serbia under the project TR-34012. Ivan Savic is a recipient of a

scholarship granted by the Ministry of Education and Science of the Republic of Serbia.

## REFERENCE

- Jin X, Zhang Y, Xiao L, Zhao Z. Optimization of extended zero-order release gliclazide tablets using D-optimal mixture design. *Journal of the Pharmaceutical Society of Japan* 2008;128:1475-83.
- Seth AK, Misra AN. Mathematical modeling of preparation of acyclovir liposomes: Reverse phase evaporation method. *Journal of Pharmacy and Pharmaceutical Sciences* 2002;5:285-91.
- Sanchez-Lafuente C, Furlanetto S, Fernandez-Arevalo M. Didanosine extended-release matrix tablets: optimization of formulation variables using statistical experimental design. *International Journal of Pharmaceutics* 2002;237:107-18.
- Guan J, Xiang R, Pan Y, Pan H, Su X, Zhou L, Cui Y, Pan W. Design and evaluation of a novel formulation prediction system. *International Journal of Pharmaceutics* 2010;402(1-2):129-39.
- Mura P, Furlanetto S, Cirri M, Maestrelli F, Marras AM, Pinzauti S. Optimization of glibenclamide tablet composition through the combined use of differential scanning calorimetry and d-optimal mixture experimental design. *Journal of Pharmaceutical and Biomedical Analysis* 2005;37:65-71.
- Chang LC, Wu SC, Tsai JW, Yu TJ, Tsai TR. Optimization of epirubicin nanoparticles using experimental design for enhanced intravesical drug delivery. *International Journal of Pharmaceutics* 2009;6(1-2):195-203.
- El-Malah Y, Nazzal S, Khanfar NM. D-optimal mixture design: optimization of ternary matrix blends for controlled zero-order drug release from oral dosage forms. *Drug Development and Industrial Pharmacy* 2006;32:1207-18.
- Rhee Y, Chang S, Park W, Chi S, Park E. Optimization of ibuprofen gel formulations using experimental design technique for enhanced transdermal penetration. *International Journal of Pharmaceutics* 2008;364(1):14-20.
- Bodea A, Leucuta SE. Optimization of hydrophilic matrix tablets using a D-optimal design. *International Journal of Pharmaceutics* 1997;153:247-55.
- Bendjaballah M, Canselier JP, Oumeddour R. Optimization of oil-in-water emulsion stability: experimental design, multiple light scattering, and acoustic attenuation spectroscopy. *Journal of Dispersion Science and Technology* 2010;31(9):1260-72.
- Marinkovic V, Karljikovic-Rajic K, Agbaba D, Nikolic M. Experimental design as a quality improvement function. *EOQ Conference*; 2004 Sept 5-7; Moscow, Russia.
- Ozil P, Rochat MH. Experimental design, an efficient tool for studying the stability of parenteral nutrition. *International Journal of Pharmaceutics* 1988;42(1-3):11-14.
- Yoshioka S, Aso Y, Uchiyama M. Statistical evaluation of non-isothermal prediction of drug stability. II. Experimental design for practical drug products. *International Journal of Pharmaceutics* 1988;46:121-32.
- Horhota ST, Burgio J, Lonski L, Rhodes CT. Effect of storage at specified temperature and humidity on properties of three directly compressible tablet formulations. *Journal of Pharmaceutical Sciences* 1976;65:1746-49.
- El-Banna HM, Ismail AA, Gadalla MAF. Factorial experiment for stability studies in the development of a tablet formulation. *Pharmazie* 1984;39:163-65.
- Dick CF, Klassen RA, Amidon GE. Determination of the sensitivity of a tablet formulation to variations in excipient levels and processing conditions using optimization techniques. *International Journal of Pharmaceutics* 1987;38:23-31.
- Bos CE. Tropical tablets: the development of tablet formulations for use in tropical countries, [dissertation]. University of Groningen; 1990.
- Bos CE, Bolhuis GK, Lerk CF. Optimization of tablet formulations based on starchflactose granulations for use in tropical countries. *Drug Development and Industrial Pharmacy* 1991;17:2373-89.
- Bos CE, Bolhuis GK, Lerk CF, De Boer JH, Duineveld CAA, Smilde AK, Doornbos DA. The use of a factorial design to evaluate the physical stability of tablets prepared by direct compression; A new approach based on the relative change in tablet parameters. *European Journal of Pharmaceutics and Biopharmaceutics* 1991;37:204-09.
- Bos CE, Bolhuis GK, Lerk CF, De Boer JH, Duineveld CAA, Smilde AK, Doornbos DA. The use of a factorial design to evaluate the physical stability of tablets prepared by direct compression; The selection of excipients suitable for use under tropical storage conditions. *European Journal of Pharmaceutics and Biopharmaceutics* 1991;37:210-15.
- Stojanovic J, Vladimirov S, Marinkovic V, Velickovic D, Sibinovic P. Monitoring of the photochemical stability of carvedilol and its degradation products by the RP-HPLC method. *Journal of the Serbian Chemical Society* 2007;72(1):37-44.
- Hokama N, Hobara N, Kameya H, Ohsiro S, Sakanashi M. Rapid and simple micro-determination of carvedilol in rat plasma by high-performance liquid chromatography. *Journal of Chromatography B* 1999;732:233-38.
- Ptacek P, Macek J, Klima J. Liquid chromatographic determination of carvedilol in human plasma. *Journal of Chromatography B* 2003;789:405-10.
- Eisenberg EJ, Patterson AM, Kahn GC. High-performance liquid chromatographic method for the simultaneous determination of the enantiomers of carvedilol and its o-desmethyl metabolite in human plasma after chiral derivatization. *Journal of Chromatography* 1989;493:105-15.
- Behn F, Laer S, Scolz H. Determination of carvedilol in human cardiac tissue by high-performance liquid chromatography. *Journal of Chromatographic Science* 2001;39(3):121-24.
- Yang E, Wang S, Kratz J, Cyronak MJ. Stereoselective analysis of carvedilol in human plasma using HPLC/MS/MS after chiral derivatization. *Journal of Pharmaceutical and Biomedical Analysis* 2004;36:609-15.

27. Saito M, Kawana J, Ohnob T, Kaneko M, Hanada K, Sugita R, Okada N, Osato S, Nagayama M, Sumiyoshi T, Ogata H. Enantioselective and highly sensitive determination of carvedilol in human plasma and whole blood after administration of the racemate using normal-phase high-performance liquid chromatography. *Journal of Chromatography B* 2006;843:73-77.
28. Lamprecht G, Gruber L, Stoschitzky K, Lindner W. Enantioselective analysis of (R)- and (S)-carvedilol in human plasma by high-performance liquid chromatography. *Chromatographia* 2002;56:S25-S30.
29. Gergov M, Robson JN, Duchoslav E, Ojanpera I. Automated liquid chromatographic/tandem mass spectrometric method for screening  $\beta$ -blocking drugs in urine. *Journal of Mass Spectrometry* 2000;35:912-18.
30. Machida M, Watanabe M, Takechi S, Kakinoki S, Nomura A. Measurement of carvedilol in plasma by high-performance liquid chromatography with electrochemical detection. *Journal of Chromatography B* 2003;798:187-91.
31. Carmo Borges NC, Mendesb GD, Silva DO, Rezende VM, Barrientos-Astigarraga RE, Nucci G. Quantification of carvedilol in human plasma by highperformance liquid chromatography coupled to electrospray tandem mass spectrometry: application to bioequivalence study. *Journal of Chromatography B* 2005;822:253-62.
32. Behn E, Laer S, Mir TS, Scholz H. HPLC quantification of carvedilol in small plasma volumes from children. *Chromatographia* 2001;53:641-44.
33. Rathod R, Prasad LP, Rani S, Nivsarkar M, Padh H. Estimation of carvedilol in human plasma by using HPLC-fluorescence detector and its application to pharmacokinetic study. *Journal of Chromatography B* 2007;857(2):219-23.
34. Zarghi A, Foroutan SM, Shafaati A, Khoddamc A. Quantification of carvedilol in human plasma by liquid chromatography using fluorescence detection: Application in pharmacokinetic studies. *Journal of Pharmaceutical and Biomedical Analysis* 2007;44:250-53.
35. Sripalakit P, Kaewnok S, Tubtonglang S. Development of carvedilol assay in tablet dosage form using HPLC with fluorescence detection. *Maejo International Journal of Science and Technology* 2010;4(01):8-19.
36. Rizwan M, Aqil M, Azeem A, Sultana Y, Talegaonkar S, Ali A. Study of the degradation kinetics of carvedilol by use of a validated stability-indicating LC method. *Chromatographia* 2009;70:1283-86.
37. ICH-Q1A (R2), Stability testing of new drug substances and products. International Conference on Harmonization: Geneva; 2003.

\*\*\*\*\*