

1,3,4-thiadiazole Attached 2, 3- disubstituted Thiazolidinones Derivatives: Synthesis and Biological Evaluation

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ABSTRACT

Background: Thiazolidinone derivatives are the subject of prominent importance since they have been seen as valuable intermediates for the formulating/synthesize of different heterocyclic derivatives and it gives various subsidiaries with every extraordinary kind of exercise. What're more, researches show that derivatives having 1,3,4-Thiadiazole core/nucleus have a wide scope of pharmacological potential that incorporates antifungal, antibacterial, antiviral, anticancer, antitubercular, anticonvulsant, antidiabetic and antioxidant. From the literature reviewed, both the nucleus was found to be active as antimicrobials, anti-inflammatory as well as anticancer agents. **Materials and Methods:** It has been planned to synthesize "1,3,4-thiadiazolyl-thiazolidin-4-one" derivatives and evaluate their biological activities. The novel subordinates (4a-4j) of 2, 3-disubstituted Thiazolidinones were integrated into acceptable yield by the synthesis of benzaldehyde with thiosemicarbazide to provide thiosemicarbazones, 1,3,4-thiadiazoles is then produced by thiosemicarbazone cyclized by involving ferric chloride. By the reaction of various aromatic aldehyde with 1,3,4-thiadiazoles yielded different Schiff bases. The final derivative 2-aryl-3-(5-aryl-1,3,4-thiadiazol-2-yl)- 1,3-thiazolidine-4-one was obtained by the reaction between the schiff base with thioglycolic acid. All subordinates/derivatives were portrayed by spectral analysis examination (IR, 1H NMR) and elemental analysis and then screened for biological activities (*in-vitro* anticancer, antimicrobial, anti-inflammatory). **Results and Conclusion:** The analysis reveals that synthesized derivatives of thiazolidinone possessing methyl, hydro, nitro, hydroxyl, fluoro, chloro, methoxy, dimethoxy and amino substitution through phenyl ring. All these groups help pharmacophore, to increase their pharmacological activities. However, the substituted phenyl ring in the side chain also facilitates the biological action of the molecules. The increased activity of the new derivatives explains that they act as a more powerful and potent bioactive molecule when compare with the standard drug. This significant activity may be due to the substitution in the 2nd position of thiazolidinone nucleus. As result, all the tested compounds exhibit good activity and compounds 4i, 4b, 4f and 4a were established more noteworthy and outstanding as of all the derivatives.

Key words: Schiff base, Thiazolidinone, 1,3,4-Thiadiazole, Molecular docking, Anti-inflammatory, Antimicrobial, Anti-cancer.

INTRODUCTION

Thiazole derivatives are the most precious classes of heterocyclic compounds because of their medicinal importance; their derivatives are characterized by high biological activity in pharmaceutical fields and have shown various pharmacological activities. Thiazolidine is a tetrahydro derivative of

thiazole and the 'oxo' derivative of thiazolidine is identified as thiazolidinone.¹ A great deal of research chip away at thiazolidinones has been done in quite a while in the past. The center/core is otherwise called magic moiety (wonder nucleus) since it gives out various derivatives with

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every single distinctive sort of pharmacological event.² Thiazolidinone subordinates are the subject of famous intrigue since they have been seen as helpful intermediates for the create/synthesize of different heterocyclic compounds.³ Thiazolidinones, which are the derivatives of thiazolidine, belongs to an essential group of heterocyclic compounds containing sulfur and nitrogen in a five-member ring.⁴ 1,3-Thiazolidin-4-ones are heterocycles that have an atom of sulfur at position 1, an atom of nitrogen at position 3 and a carbonyl group at position 4. Replacement can occur at 2, 3 and 5 positions.² For the planning of potential bioactive, a Thiazolidin-4-one an adaptable platform. Thiazolidin-4-one compounds accounted for an expansive range of pharmacological activity, for example anticancer, cell reinforcement, antimicrobial,⁶ anti-inflammatory,⁵ and anti-HIV,⁷ anticonvulsant,⁸ and cardioprotective activities.⁹

Due to this bioactivity and pharmacological impact of the compound and in the persistence of our ongoing study on heterocyclic compounds, we decided to synthesize a novel series of thiazolidin-4-one” derivatives, in which 1,3,4-Thiadiazole nucleus was added as intermediate. Writing demonstrates that a molecule having 1,3,4-Thiadiazole core has a wide scope of pharmacological exercises that incorporate antifungal,¹⁰ antibacterial,¹¹ antiviral, anti-inflammatory,¹² antitubercular,^{13,14} pain-relieving, antileishmanial, antiepileptic, CNS depressant, anticancer,¹⁵ anticonvulsant, cell reinforcement, diuretic,¹⁶ molluscicidal, antidiabetic,¹⁷ and antihypertensive.¹⁸

From the scientific works, both the nucleus was found to be active as anticancer, antimicrobials and anti-inflammatory agents. It has been planned to synthesize “1,3,4-thiadiazolyl-thiazolidin-4-one” derivatives. The different Schiff bases were yielded on the treatment of 2-amino-5-aryl 1,3,4-thiadiazoles with aromatic aldehydes, which were cyclized by reaction with thioglycolic acid using *N,N'*-Dicyclohexylcarbodiimide (DCC) as dehydrating agents and evaluate them for their biological activity (i.e. anticancer, antimicrobials and anti-inflammatory agents).

Experimental

By utilizing the TLC plate method, the compound response was identified and evaluated by introducing benzene-ethanol (8:2) as a solvent system and applying ultraviolet light on the TLC plate, the compounds were spotted/identified in presence of iodine fumes. By utilizing digital melting point apparatus, the melting points of the various synthesized derivatives were identified. The FTIR method (Cary-60, Agilent Technologies, USA) was employed to characterize the

various groups (i.e. C=O, NH spectra etc.). Moreover, the other essential characteristics of the prepared synthesized compounds were characterized using NMR spectral estimation involving ¹H and ¹³C NMR spectra (Bruker DPX-300 MHz), CDCl₃ was utilized as a solvent system in this spectral analysis and Mass spectra (LCMS/MS-Applied Bio-system, USA). Rather then, the other essential chemicals involved in the duration of the synthesis procedure were purchased from Sigma Aldrich, USA; Himedia Laboratory (MS), India.

Synthesis of Thiosemicarbozone

0.2 M Aromatic aldehyde was dissolved in 300 ml of warm ethanol. 0.2M of Thiosemicarbazide was dissolved in 300 ml of hot water. Both the solutions were mixed slowly with continuous stirring. After cooling, the separated crystals of the product (thiosemicarbozone) were filtered off. The purity of the compound was monitored by TLC and melting point determination.¹⁹

Amalgamation of 2-Amino-5-Aryl-1,3,4-Thiadiazoles

0.05M of Thiosemicarbozone was suspended in 300 ml of double-distilled water in a beaker. Ferric chloride (0.15M), disintegrated in 300 ml of (distilled water) DW and then included in thiosemicarbozone dispersion. It was warmed to 80-90°C and kept up for 45 min and afterward, the dispersion was filtered. A blend of citrus acid (0.11M) and sodium citrate (0.05M) was added to the above dispersion and mixed gently using a homogenizer. Subsequent to cooling, the prepared reaction was neutralized with 10% of the ammonia solution. The precipitate then acquired was separated and washed with DW and permitted to dry then recrystallized by using 40% ethanol.²⁰

Synthesis of Schiff Bases

0.02M of the amino-thiadiazole was dissolved in 10 ml of absolute ethanol by heating and stirring. Then, 0.02M appropriate aldehyde in 10 ml ethanol was added to it, with stirring. Then refluxed it for 3-4 hr. The crystals obtained on cooling were filtered and dried. The purity of the compounds were monitored by TLC and melting point determination.²¹

Synthesis of 4-Thiazolidinone Derivatives

0.01M of the Schiff base was included in 100 ml of benzene, subsequently, 0.01M thioglycolic acid and 0.01M dicyclohexylcarbodiimide (DCC) was included. The dispersion was refluxed for 6 hr. At that point the concentrated the above dispersion utilizing vacuum evaporation. The concentrate was neutralized by the addition of NaHCO₃ solution (10%) and the precipitate acquired was separated by filtration and dried.²²⁻²⁴

Spectral analysis of synthesized compound**Thiosemicarbazone(I)**

IR- 1267 (C=S), 1452 (C=C Stretch), 1520 (C=N Stretch), 3060(C-H Stretch). ¹H NMR- 2.0, 8.1, 7.6, 7.3. ¹³C NMR- 181.4, 143.0, 133.8, 129.2.m/e: 179.5. Elemental analysis (C₈H₉N₃S): N, 23.44; S, 17.89, H, 5.06; C, 53.61; found N:22.53, H:4.68, C:52.45.

1,3,4-thiadiazole (II)

IR- 1740(C=O), 1634 (C=N), 1423(C=C) 3111(C-H). ¹H NMR- 7.48, 7.32, 7.22, 4.0. ¹³C NMR- 161.6, 127.5, 129.3, 128.8, 133.5.m/e: 193.2. Elemental analysis (C₉H₁₁N₃S): N, 21.74; H, 5.74; C, 55.93, found N:20.16, H: 3.11, C: 54.83,

2-phenyl-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one.(4a)

IR- 1636, 1458 (C=C)1663 (C=O), 1544 (C=N Stretch), 3134 (C-H Stretch). ¹H NMR- 5.92, 7.06, 7.14, 7.48, 7.32. ¹³C NMR- 127.2, 128.7, 139.2, 171.2, 65.3, 33.6, 163.4, 175.0, 133.5.m/e: 339.5. Elemental analysis (C₁₇H₁₃N₃O₂S): C: 61.45, H:3.53, N:12.53, found C:60.15, H:3.86, N:12.38.

2-(4-(dimethylamino)phenyl)-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one.(4b)

IR- 1610, 1458 (C=C Stretch), 1217, 1284 (N-CH₃ Stretch), 1683 (C=O Stretch), 1521 (C=N Stretch), 3018 (C-H Stretch). ¹H NMR- 2.85, 6.47, 6.88, 5.92, 7.48, 7.32. ¹³C NMR- 40.3, 148.0, 114.2, 129.7, 128.7, 65.3, 171.2, 163.4, 175.0, 127.5, 129.3.m/e: 382.9; Elemental analysis(C₁₉H₁₈N₄O₂S): C: 60.45, H:4.53, N:13.53, found C:59.66, H: 4.74, N:14.65.

2-(2, 3-dimethoxyphenyl)-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one(4c)

IR- 1319, 1280 (C-O Stretch), 3062 (C-H Stretch), 1627, 1442 (C=C Stretch), 1666 (C=O Stretch),1517 (C=N Stretch), ¹H NMR- 3.73, 6.47, 6.59, 6.51, 5.92, 7.48, 7.32, 7.22. ¹³C NMR- 56.2, 56.5, 149.7, 113.7, 150.2, 117.6, 163.4, 175.0, 133.5, 128.8.m/e: 399.07; Elemental analysis (C₁₉H₁₇N₃O₃S₂): C: 57.45, H:4.53, N:11.03, found C: 57.12, H: 4.29, N:10.52.

2-(4-chlorophenyl)-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one(4d)

IR- 1045 (C-Cl Stretch),3066 (C-H Stretch), 1616, 1444 (C=C Stretch), 1699 (C=O Stretch), 1527 (C=N Stretch). ¹H NMR- 7.15, 7.00, 5.92, 3.33, 7.38, 7.32, 7.22. ¹³C NMR- 132.7, 128.8, 130.2, 137.3, 65.3, 171.2, 163.4, 175.0, 133.5, 127.5, 129.3, 128.8. m/e: 373.1, Elemental analysis(C₁₇H₁₂ClN₃O₂S): C: 54.95, H:3.82, N:11.53, found C: 53.61, H: 3.24, N:11.24.

2-(4-nitrophenyl)-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one.(4e)

IR-, 1544, 1376 (C-NO₂ Stretch), 1508 (C=N Stretch), 3132 (C-H Stretch), 1608, 1409 (C=C Stretch), 1637 (C=O Stretch). ¹H NMR- 8.07, 7.32, 5.92, 3.38, 7.48, 7.32,7.22. ¹³C NMR- 146.8, 121.0, 129.7, 145.3, 171.2, 163.4, 175.0, 127.5.m/e: 384.04, Elemental analysis (C₁₇H₁₂N₄O₃S₂):N:15.03, H:4.53, C: 53.45, found N:14.57, H: 3.15, C:53.11.

2-(4-hydroxyphenyl)-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one.(4f)

IR- 3132 (C-H Stretch), 1601, 1412 (C=C Stretch), 3460(O-H Stretch), 1520 (C=N Stretch), 1640 (C=O Stretch). ¹H NMR- 6.61, 6.89, 5.92, 7.48, 7.32, 7.22. ¹³C NMR- 115.8, 130.2, 131.8, 130.2, 171.2, 163.4, 175.0, 128.8, 129.3.m/e: 355.04, Elemental analysis (C₁₇H₁₃N₃O₂S₂):C: 56.65, H:3.91, N:12.58, found C: 57.45, H: 3.69, N: 11.82.

2-(4-methoxyphenyl)-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one(4g)

IR- 1300 (Ar-C-N Stretch), 1512 (C=N Stretch),1623 (C=C Stretch), 3145 (C-H Stretch), 1647 (C=O Stretch). ¹H NMR- 3.73, 6.65, 6.95, 5.92, 7.48, 7.32, 7.22.¹³C NMR- 55.9, 159.1, 114.2, 129.8, 131.5, 33.6, 171.2, 65.3, 163.4. m/e: 369.6. Elemental analysis (C₁₈H₁₅N₃O₂S₂): C: 57.45, H:3.22, N:12.43, found C: 58.52, H: 4.09, N: 11.37.

3-(5-phenyl-1,3,4-thiadiazol-2-yl)-2-p-tylthiazolidin-4-one.(4h)

IR- 1505 (C=N Stretch), 1656, 1432 (C=C Stretch), 1631 (C=O Stretch), 3130 (C-H Stretch),. ¹H NMR- 6.94, 5.92, 3.38, 7.48, 7.32, 7.22. ¹³C NMR- 136.8, 129.0, 128.7, 136.2, 65.3, 171.2, 163.4, 175.0, 127.5, 129.3, 128.8.m/e: 353.7. Elemental analysis (C₁₈H₁₅N₃O₂S₂): C: 62.45, H:5.53, N:11.33, found C: 61.16, H: 4.28, N:11.89.

2-(4-aminophenyl)-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one.(4i)

IR- 3152(N-H stretch), 1510 (C=N Stretch), 1645, 1409 (C=C Stretch), 3136 (C-H Stretch), 1622 (C=O Stretch). ¹H NMR- 6.34, 6.81, 5.92, 7.48, 7.32, 7.22, 7.48. ¹³C NMR- 129.6, 116.2, 146.8, 116.2, 163.4, 171.2, 175.0, 127.5, 133.5, 128.3.m/e: 354.06, Elemental analysis (C₁₇H₁₄N₄O₂S₂):C: 56.45, H:4.67, N:14.53, found C: 57.61, H: 3.98, N:15.81.

2-(4-fluorophenyl)-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one(4j)

IR- 945(Ar-C-F Stretch), 1501 (C=N Stretch), 3115 (C-H Stretch),1672, 1410 (C=C Stretch), 1627 (C=O Stretch). ¹H NMR- 6.85, 7.04, 5.92, 7.48, 7.32. ¹³C NMR- 161.3,

115.4, 130.4, 134.8, 171.2, 163.4, 133.5, 127.5, 129.3, 128.8.m/e: 357.4. Elemental analysis (C₁₇H₁₂FN₃OS₂):C: 57.12, H:3.53, N:11.53, found C: 57.43, H: 3.38, N:11.36.

Biological evaluation

In vitro anticancer activity

The SRB-based anticancer assay is an economical, easy and sensitive process to identify the cytotoxic capability of the test molecule. To get a large amount of cells, the acquired human breast cancer cells were grown, containing a suitable culture medium in 5% CO₂ with RH (relative humidity) 90% and temperature (37°C). After grown of cells, the cell was treated with a trypsin-EDTA solution and maintained the growth to 10,000 cells/100 µl in suspension, then the adequate amount of cell suspension (100 µl) was taken in 96 well plates and incubated at the atmospheric condition mentioned above (37°C, 5% CO₂ and RH 90%) for 24 hr. After then the different concentration of the test sample (10, 20, 40 and 80 microgram/ml) was added individually in a separate tube. Trichloric acid (50 µl-50% chilled solution) was then added into the well plate after (48 h) addition of sample. For fixation of cells, the plates were again incubated at 4°C for 1 h. Before the addition of SRB solution (100 µl SRB solution of 0.4%w/v in 1% acetic acid), the plates were washed and air dried, then the plates were placed for 30 min at room temperature and then washed with acetic acid solution (1%) followed by air-dried. Tris buffer (100 µl of 10.5 M) was then added into the well plates and placed into a mechanical shaker for shaking for about 20 mins. After all the above process the cells (cell growth inhibition) was identified and evaluated by using ELISA reader for about 540nm.^{25,26}

Antimicrobial assessment

By comparison with standard drugs ampicillin and miconazole, the prepared series (4a-4j) was found to be active as an antibacterial and antifungal moiety.²⁷

Evaluation of Antibacterial activity

By utilizing various concentration range (0.025µg/ml to 2500µg/ml) of the synthesized derivatives (test compound) was preferred to characterize the antibacterial assay by agar well diffusion technique on Mueller Hinton agar media (bacterial nutrient medium) against *Staphylococcus aureus* (ATCC 10231), *Escherichia coli* (ATCC25922), *Streptococcus pyogenes* (MTCC 442) and *Pseudomonas aeruginosa* (ATCC 10145), DMSO utilized as control and Ampicillin drug was used as a standard against Gram-positive and Gram-negative bacteria.

Evaluation of Antifungal Activity

The synthesized compounds (4a-4j) were screened for antifungal activity, at a different concentration of (0.025µg/ml to 2500µg/ml). The antifungal evaluation of the synthesized derivative was identified using Sabouraud dextrose media (fungal nutrient medium). DMSO was utilized as a control medium and Miconazole was used as a standard against *Candida albicans* ATCC 24433 and *Aspergillusniger*, ATCC 16888.

Anti-inflammatory assay

The anti-inflammatory potential of integrated mixes was assessed via the carragenan-prompted hind paw edema technique.²⁸ The Institutional Animal Ethics Committee approval number for performing *in-vivo* assay is GRKIST/406/02/IAEC/16A. Ibuprofen was utilized as a source of perspective medication. The test was performed in the rat (using an animal model) and then the animal was separated into three different groups having 6 rats in each group. Group I was served as Control, II group was treated with the test sample, whereas the III group was utilized for standard treatment. In each animal, the aponeurosis was grown of the right hind paw by applied dispersion of carrageenan (1% in 0.9% saline- 0.05 ml). One hour before treatment of carrageenan, the test sample and standard formulation were administered orally into the treated and standard group respectively. Before 1 hr and after 3 hr of the carrageenan administration, the paw volume of each animal was estimated by utilizing a plethysmometer. The % anti-inflammatory potential of the synthesized derivatives was determined by the equation shown beneath:

$$\% \text{ inhibition of edema} = (1 - V_t / V_c) \times 100$$

Here, the mean enhance in paw volume of rat in treated group represents as V_t and for control group represented by V_c, respectively.

Computational chemistry studies

Molecular Docking study

Molecular docking analysis was performed by utilizing the LibDock module in Discovery Studio 2.1, to examine the inhibitory activity and orientation of ligand against human DHFR. To estimate the anticancer activity, firstly, the synthesized compound was docked with the 3D structure of human DHFR (retrieved from the Protein Data Bank- PDB ID: 1DLS;http://www.rcsb.org) and afterward with the identified anticancer molecule (drug) Methotrexate. The pre-owned docking program LibDock produces a few represents, each

delivering their relating LibDock scores with various directions inside the characterized effective site of the protein. The best ligand binding verification was characterized by an optimized high LibDock score of the ligand pose. At long last, the examine ligand (synthesized compound) poses subprotocol was performed to estimate hydrogen bonds and hydrogen bond distance (close contacts-Vander Waals conflicts between the test sample and human DHFR.

ADMET prediction

The compounds are analyzed by their respective toxicity and pharmacokinetic studies by utilizing the ADMET descriptors analysis protocol in DS.

RESULTS AND DISCUSSIONS

In our incessant effort to synthesize some novel thiazolidinone compounds active as anticancer, anti-microbial and anti-inflammatory agents, we report a series of thiadiazolythiazolidinone derivatives. Inspection of the chemical structure of the target compounds revealed that the nucleus could be divided into two subunits: the thiazolidinone part and the thiadiazole part. The TLC analysis was a utility to characterize the purity of the compounds and the recrystallization of the compound was done utilizing the solvent-ethanol. The spectral assay NMR and IR was employed to characterize the spectral features of the various synthesized derivatives as well as the elemental analysis also been employed.

Chemistry

The scheme of synthesis of different thiazolidinone derivative shown in Figure 1. The synthesized derivatives of thiazolidinone possessing methyl, hydro, nitro, hydroxyl, fluoro, chloro, methoxy, dimethoxy and amino substitution through phenyl ring (Table 1). All these groups help pharmacophore, to increase their pharmacological activities. However, the substituted phenyl ring in the side chain also facilitates the biological action of the molecules.²⁹

In their similar investigation of the substituted group of the aryl moiety towards antibacterial potency, they observe that more activity was found in electron-withdrawing moiety and less activity was shown in the group having electrons donating moiety.³⁰ Effect of electron-withdrawing groups present in the synthesized compounds on the result, is greater than our expectations. Our present study shows the importance of the effect of substitutions on the thiazolidinone moiety for their pharmacological properties.

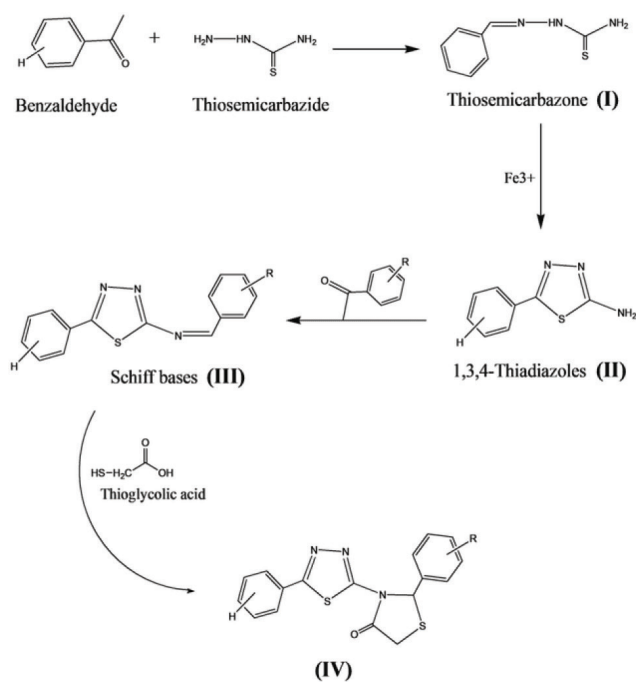


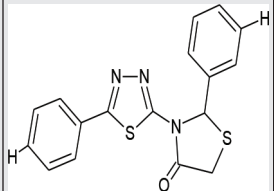
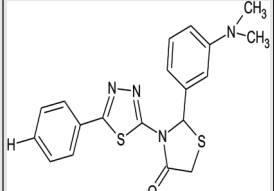
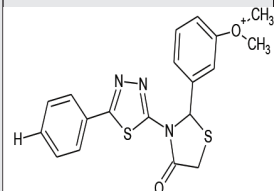
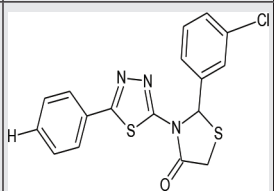
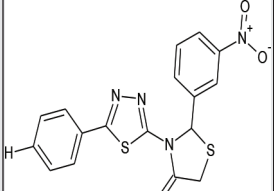
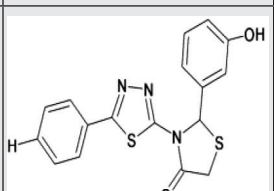
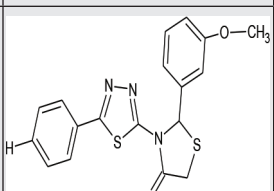
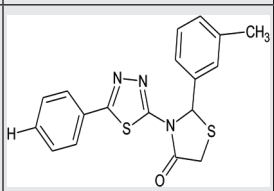
Figure 1: Scheme of synthesis.

In systems containing more than two hetero atoms in the same ring, the trend of properties are found. In particular, the additional heteroatoms in both six- and five-membered systems leads to a suppression of electrophilic substitution and a slowing of electrophilic addition to nitrogen. On the other hand, further increase in tendencies for nucleophilic substitution and addition and the five-membered compounds, further increases in acidities of N-hydrogen are found.³¹ Thiadiazoles are extremely weak bases because of the inductive impacts of the extra hetero molecules, despite the fact that N-quaternized responses can be done. For comparative reasons, electrophilic replacements on carbon are for all intents and purposes obscure, aside from mercurations and halogenations.³²

In-vitro anticancer activity

The human breast cancer cell lines (MCF-7) were utilized to predict the anticancer potential of the synthesized thiazolidinone derivatives (4a-4j) by SRB assay. The cytotoxicity assessment suggests the dose-dependent manner of the system against cancer cells which justifies the decrement in cell viability by enhancing the sample (synthesized derivative) concentration. The outcome revealed by the assay was depicted in the graph (Figure 2), which suggested the higher concentration of synthesized derivative shows greater inhibitory action. Different derivatives (4a-4j) were observed to display enhanced cytotoxic effect with increasing the concentration range from 10, 20, 40 and 80µg/ml. The

Table 1: Physicochemical properties of synthesized derivatives.

S. No.	Code	R	Compound	Molecular formula	Mol. Weight	Yield (%)	M.P. (°c)	R _f
1.	4a	H		C ₁₇ H ₁₃ N ₃ OS ₂	339	44.52	527.3	0.62
2.	4b	N(CH ₃) ₂		C ₁₉ H ₁₈ N ₄ OS ₂	382	38.22	594.8	0.55
3.	4c	(OCH ₃) ₂		C ₁₉ H ₁₇ N ₃ O ₃ S ₂	399	46.43	521.2	0.72
4.	4d	Cl		C ₁₇ H ₁₂ ClN ₃ OS ₂	373	38.17	569.7	0.64
5.	4e	NO ₂		C ₁₇ H ₁₂ N ₄ O ₃ S ₂	384	35.86	539.1	0.59
6.	4f	OH		C ₁₇ H ₁₃ N ₃ O ₂ S ₂	355	30.45	509.5	0.66
7.	4g	OCH ₃		C ₁₈ H ₁₅ N ₃ O ₂ S ₂	369	32.34	573.3	0.67
8.	4h	CH ₃		C ₁₈ H ₁₅ N ₃ OS ₂	353	38.62	551.1	0.71

Continued...

Table 1: Cont'd.

S. No.	Code	R	Compound	Molecular formula	Mol. Weight	Yield (%)	M.P. (°c)	R _f
9.	4i	NH ₂		C ₁₇ H ₁₄ N ₄ OS ₂	354	33.75	623.1	0.70
10.	4j	F		C ₁₇ H ₁₂ FN ₃ OS ₂	357	28.73	540.4	0.71

cytotoxic effect of different derivatives (4a-4j) in MCF-7 was exposed to have superior inhibitory potential. Anticancer assessment of various derivatives results indicated that compounds 2-(4-aminophenyl)-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one (4i), 2-(4-(dimethylamino)phenyl)-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one (4b), 2-(4-hydroxyphenyl)-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one (4f) and 2-phenyl-3-(5-phenyl-1,3,4-thiadiazol-2-yl)thiazolidin-4-one (4a) were the most intense cytotoxic specialists. According to obtained results, it was concluded that the most specific and effective activity of the derivatives, found on those derivative having groups like amine, hydrogen moiety, hydroxyl group at the phenyl ring of thiazolidin-4-one, as the compare with others. Similarly, other derivatives having following functional moiety i.e. NO₂, Chloro, fluoro, CH₃ also suggested anticancer activity.^{5,33}

Antimicrobial activity

Thiadiazole system in the side chain linked to the thiazolidinone moiety helps, (4a-4j), the compounds to show antimicrobial activity, at a concentration value of 0.025-2500 µg/ml. The two sections (the thiazolidinone part and the thiadiazole part) have been accounted for to have a huge wide range of antimicrobial exercises which may add to the great outcomes got from testing them as antibacterial operations. In addition, replacement of the distal phenyl aromatic ring from thiazolidinone moieties with various grouping at the second position may likewise help in getting such great outcomes.^{34,35} As our result shows all the compounds are active and show antimicrobial against all the strains. It was interesting to observe that some compounds (4a, 4b, 4f and 4i) have shown activity that is very near to the standard drugs (shown in Table 2 and 3).

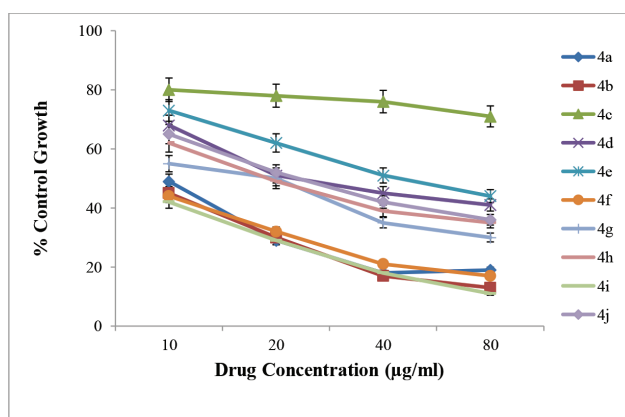


Figure 2: In-vitro anticancer study of different derivatives in MCF-7 cancer cell line.

The increased activity of the new derivatives can be explained that they act as more powerful and potent microbicidal agents, thus killing more of the bacteria and fungi than the standard drug. This is might be the result of the lipophilic character, which favors its permeation through the lipid layer of the microbial membranes. The existence of imino linkage (-N=C-) in the compounds is essential for the enhancement of antibacterial and antimicrobial activities.³⁶

Anti-inflammatory activity

Analogues also show encouraging results against anti-inflammatory activity. All the compounds (4a-4j) were screened for anti-inflammatory activity, using ibuprofen as a reference drug. All the tested compounds exhibit good activity. Compound 4i, which was substituted by NH₂ group at the C-2 position of thiazolidin-4(3H)-one moiety is the most active compound, among all the tested compounds showed 58.64% potency and

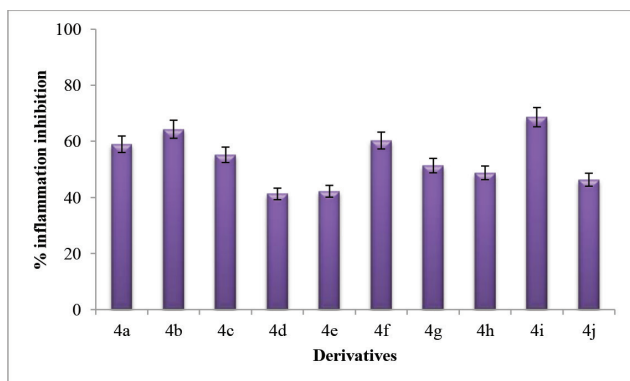
Table 2: Anti-bacterial bio-assay of synthesized compounds.

Compounds	Bacteria 2500		Diameter of the zone (in mm) as per Concentration Placed in agar well (in µg/ml)					
			250	25	2.5	0.25	0.025	
4a.	Gram-ve	<i>P. aeruginosa</i>	22	15	11	-	-	-
		<i>E. coli</i>	20	13	10			
	Gram+ve	<i>S. aureus</i>	21	16	11	-	-	-
		<i>S. pyogenes</i>	20	16	12			
4b.	Gram-ve	<i>P. aeruginosa</i>	26	22	16	13	-	-
		<i>E. coli</i>	22	15	10		-	-
	Gram+ve	<i>S. aureus</i>	23	17	10	-	-	-
		<i>S. pyogenes</i>	24	19	11	-	-	-
4c.	Gram-ve	<i>P. aeruginosa</i>	22	17	11	-	-	-
		<i>E. coli</i>	19	13	10	-	-	-
	Gram+ve	<i>S. aureus</i>	20	15	11	-	-	-
		<i>S. pyogenes</i>	21	15	12	-	-	-
4d.	Gram-ve	<i>P. aeruginosa</i>	21	17	12	-	-	-
		<i>E. coli</i>	18	16	9	-	-	-
	Gram+ve	<i>S. aureus</i>	17	14	9	-	-	-
		<i>S. pyogenes</i>	17	13	8	-	-	-
4e.	Gram-ve	<i>P.aeruginosa</i>	16	12	9	-	-	-
		<i>E. coli</i>	15	10	9	-	-	-
	Gram+ve	<i>S. aureus</i>	14	10	10	-	-	-
		<i>S. pyogenes</i>	15	9	8	-	-	-
4f.	Gram-ve	<i>P.aeruginosa</i>	23	17	15	9	-	-
		<i>E. coli</i>	22	14	12	-	-	-
	Gram+ve	<i>S. aureus</i>	21	15	11	-	-	-
		<i>S. pyogenes</i>	23	15	10	6	-	-
4g.	Gram-ve	<i>P. aeruginosa</i>	20	14	10	-	-	-
		<i>E. coli</i>	20	13	9	-	-	-
	Gram+ve	<i>S. aureus</i>	19	14	-	-	-	-
		<i>S. pyogenes</i>	21	15	11	-	-	-
4h.	Gram-ve	<i>P. aeruginosa</i>	19	14	10	-	-	-
		<i>E. coli</i>	19	12	8	-	-	-
	Gram+ve	<i>S. aureus</i>	20	15	11	-	-	-
		<i>S. pyogenes</i>	18	12	-	-	-	-
4i.	Gram-ve	<i>P. aeruginosa</i>	26	20	18	9	-	-
		<i>E. coli</i>	26	18	12	-	-	-
	Gram+ve	<i>S. aureus</i>	24	16	11	-	-	-
		<i>S. pyogenes</i>	25	17	12	7	-	-
4j.	Gram-ve	<i>P. aeruginosa</i>	19	14	11	-	-	-
		<i>E. coli</i>	18	15	9	-	-	-
	Gram+ve	<i>S. aureus</i>	20	14	10	-	-	-
		<i>S. pyogenes</i>	19	14	11	-	-	-
Ampiciline	Gram-ve	<i>P. aeruginosa</i>	26	21	17	12	-	-
		<i>E. coli</i>	24	19	10	-	-	-
	Gram+ve	<i>S. aureus</i>	25	19	11	-	-	-
		<i>S. pyogenes</i>	25	18	12	8	-	-

Table 3:Anti-fungal activity of synthesized compound.

Compounds	Fungi	Diameter of zone (in mm) as per Concentration Placed in agar well (in µg/ml)					
		2500	250	25	2.5	0.25	0.025
4a.	<i>A. niger</i>	21	16	11	-	-	-
	<i>C. albicans</i>	22	17	13	8	-	-
4b.	<i>A. niger</i>	24	20	15	10	-	-
	<i>C. albicans</i>	23	18	12	-	-	-
4c.	<i>A. niger</i>	22	15	-	-	-	-
	<i>C. albicans</i>	17	14	10	-	-	-
4d	<i>A. niger</i>	22	15	-	-	-	-
	<i>C. albicans</i>	17	14	10	-	-	-
4e.	<i>A. niger</i>	19	15	10	-	-	-
	<i>C. albicans</i>	21	16	11	-	-	-
4f.	<i>A. niger</i>	22	16	12	-	-	-
	<i>C. albicans</i>	20	15	12	-	-	-
4g.	<i>A. niger</i>	20	18	11	-	-	-
	<i>C. albicans</i>	22	17	13	-	-	-
4h.	<i>A. niger</i>	20	17	12	-	-	-
	<i>C. albicans</i>	18	14	10	-	-	-
4i.	<i>A. niger</i>	25	19	13	8	-	-
	<i>C. albicans</i>	23	17	12	-	-	-
4j.	<i>A. niger</i>	20	14	10	-	-	-
	<i>C. albicans</i>	23	17	11	-	-	-
Miconazole	<i>A. niger</i>	25	19	14	-	-	-
	<i>C. albicans</i>	22	17	14	10	-	-

(- = not active)

**Figure 3: Anti-inflammatory activity of different derivatives.**

Compound 4d, exhibit lower percentage inhibition of edema which is 41.27% (as shown in Figure 3). The significant activity may be due to the presence of electron-withdrawing group in the 2nd position of thiazolidinone nucleus.³⁷

Molecular docking study

The data obtained from the docking study between ligand and protein molecule was depicted in Table 4 and

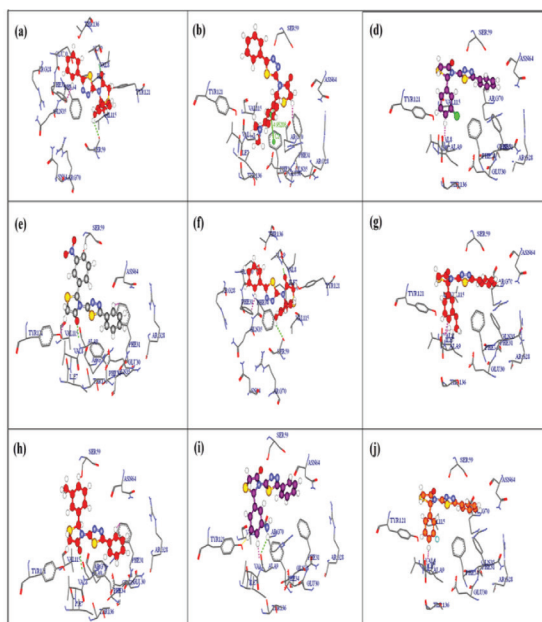
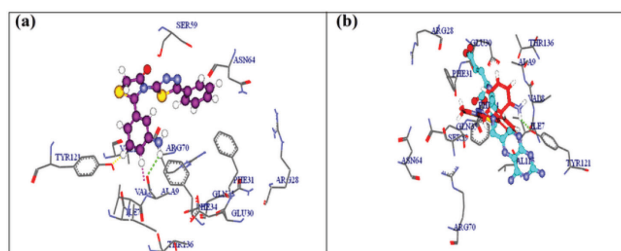
also illustration was summarized in Figure 4. According to the above results, it was found that the compound synthesized derivative 4i, signifies the effective binding confirmation (score- 128.488 kcal/mol) with highly negative bonding charges (-28.09) with the human DHFR, which is near to Methotrexate 168 K.cal/mol, the docking pattern also revealed the compound 4i fitted well in the active site pocket. As the study, among other derivatives, also possess their significant binding score and their active participation with DHFR. From the derivative 4i, the three hydrogen (H) bond was formed with DHFR, the first hydrogen bond is formed with the amino acid ALA9 of the DHFR with the 38th hydrogen of derivative 4i (4i:H38 - A:ALA9:O) with H-distance 2.47400 Å⁰. In the next two more H-bond was formed with ALA9 (A:ALA9:HN - 4i:H36) and TYR (4i:H35 - A:TYR121:OH) with a hydrogen bond distance 1.63900 Å⁰ and 1.91300 Å⁰ respectively. (Figure 5). Interaction of compound 4c was not been obtained, for that, the docking data and Figure not been represented.

ADMET study

The results are determined by utilizing standard parameters provided by DS. The results are given in Table 5

Table 4: Docking scores of the compounds (A1 4a-4J) with human DHFR (PDB ID:1DLF) in comparison with Methotrexate) docking score.

Name	Libdock Score	Interacting Amino Acids	Binding Energy	Interacting Atoms	H-Distance
4f	119.714	ILE7, Val8, Ala9, Tyr121, Val115, Arg70, Asn64, SER59, Phe34, Phe31, Glu30, Gln35, Thr136	-14.73122	4f:H37 - A:SER59:OG A:ALA9:HN - 4f:O17 A:PHE31:CZ - 4f:H25	2.476000 2.164000 2.104000
4e	127.686	ILE7, Val8, Ala9, Tyr121, Val115, Arg70, Asn64, SER59, Phe34, Phe31, Glu30, Gln35, Thr136	37.90698	A:ALA9:HN - 4e:O17 4e:H27 - A:PHE31:CZ 4e:H34 - A:TYR121:OH	1.967000 2.142000 1.973000
4h	117.664	ILE7, Val8, Ala9, Tyr121, Val115, Arg70, Asn64, SER59, Phe34, Phe31, Glu30, Gln35, Thr136	-4.98231	A:ALA9:HN - 4h:O17 4h:H25 - A:PHE31:CZ	2.198000 2.116000
4i	128.488	ILE7, Val8, Ala9, Tyr121, Val115, Arg70, Asn64, SER59, Phe34, Phe31, Glu30, Gln35, Thr136	-28.09996	4i:H38 - A:ALA9:O A:ALA9:HN - 4i:H36 4i:H35 - A:TYR121:OH	2.474000 1.639000 1.913000
4b	127.381	ILE7, Val8, Ala9, Tyr121, Val115, Arg70, Asn64, SER59, Phe34, Phe31, Glu30, Gln35, Thr136	-17.31804	4b:H34 - A:PHE31:CE1	2.165000
4a	119.185	ILE7, Val8, Ala9, Tyr121, Val115, Arg70, Asn64, SER59, Phe34, Phe31, Glu30, Gln35, Thr136	-12.57253	4a:H34 - A:TYR121:OH	1.961000
4c	127.345	ILE7, Val8, Ala9, Tyr121, Val115, Arg70, Asn64, SER59, Phe34, Phe31, Glu30, Gln35, Thr136	-24.27834	No interactions	
4d	125.194	ILE7, Val8, Ala9, Tyr121, Val115, Arg70, Asn64, SER59, Phe34, Phe31, Glu30, Gln35, Thr136	11.15526	A:ALA9:HN - 4d:H36	1.595000
4g	128.106	ILE7, Val8, Ala9, Tyr121, Val115, Arg70, Asn64, SER59, Phe34, Phe31, Glu30, Gln35, Thr136	-4.98231	A:ALA9:HN - 4g:C23 A:ALA9:HN - 4g:H37	2.173000 1.302000
4j	126.234	ILE7, Val8, Ala9, Tyr121, Val115, Arg70, Asn64, SER59, Phe34, Phe31, Glu30, Gln35, Thr136	-2.62483	4j:H36 - A:ALA9:HN	1.556000

**Figure 4: Receptor-ligand Hydrogen bonding interactions of 4(a-j) compound with active site residues of human DHFR.****Figure 5: (a) Receptor-ligand Hydrogen bonding interactions of 4i compound with active site residues of human DHFR (Green dotted lines-Hydrogen bonds, Pink Colour-Bumps); (b) Superimposition of methotrexate with 4i compound in active site pocket (methotrexate represented as Ball-stick and Stick mode 4i).**

and Figure 6. As per DS parameters, standard values like level 0 for human intestinal absorption, 3 and 4 for the solubility. While level 0 for non-inhibitory feature with CYP450 2D6, 3 for the BBB penetration and 0 for the non-toxicity nature were checked for acquiring drug substances. Figure 6 displays the predicted values

Table 5: Predicted ADMET properties of the compounds.

Compound	ADMET BBB Level	ADMET Absorption Level	ADMET Solubility Level	ADMET Hepato-toxicity	ADMET CYP2D6	ADMET PPB Level	ADMET AlogP98
4a	1	0	2	1	0	2	4.104
4b	1	0	2	0	1	2	4.266
4c	1	0	2	0	1	2	4.158
4d	1	0	1	1	0	2	4.769
4e	2	0	2	1	0	2	3.999
4f	1	0	2	1	0	2	3.862
4g	1	0	2	1	1	2	4.088
4h	1	0	2	1	0	2	4.59
4i	2	0	2	1	0	2	3.358
4j	1	0	2	1	0	2	4.31

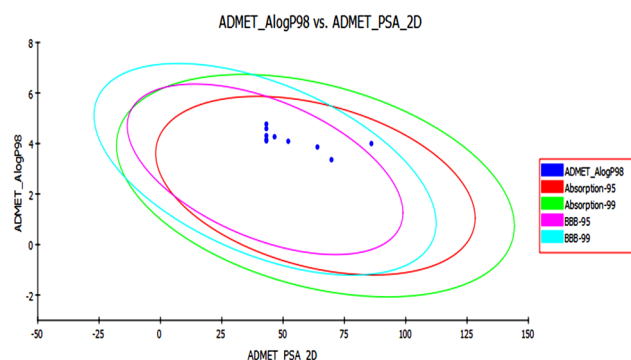


Figure 6: Plot of PSA vs Log.

plot of drug absorption for the compounds. The X-axis denotes the compound's solubility and Y-axis denotes the log P values. ADMET Descriptors, the 2D polar surface area in A2 is plotted against ALogP98. The space occupied by the ellipse indicates absorption without violating ADMET features. Ellipses denote the model at 95-99% confidence limit to the absorption models of BBB and intestinal.

CONCLUSION

Most of the therapeutic agents are heterocyclic compounds; hence heterocyclic chemistry has been the most fruitful area for drug discovery. Based on this fact, the present investigation was planned. According to the literature survey it found that thiazolidinone compound have wide varieties of biological activities like hypnotic, antitussive, analgesic, muscle relaxant, anticonvulsant, antiallergic, antibacterial, hypoglycemic and anti-spasmodic. So, considerable interest has been found in thiazolidinone compounds in search of potential

drugs. Small changes in thiazolidinone can convert it into 2,3-disubstituted thiazolidinone and the biological activity can be enhanced with the introduction of the 1,3,4-thiadiazole. According to results, the derivatives of the principle compound can be synthesized in a stable and solid crystalline form. The synthesized derivatives were then confirmed by FT-IR, Melting point and NMR spectrometer and screened for biological evaluation. All the derivatives are effective as bioactive agent. Overall, this study provides sound evidence that our synthesized compounds can be converted and then use as a suitable dosage form because they possess favorable characteristics amenable to further development towards commercial application.

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CONFLICT OF INTEREST

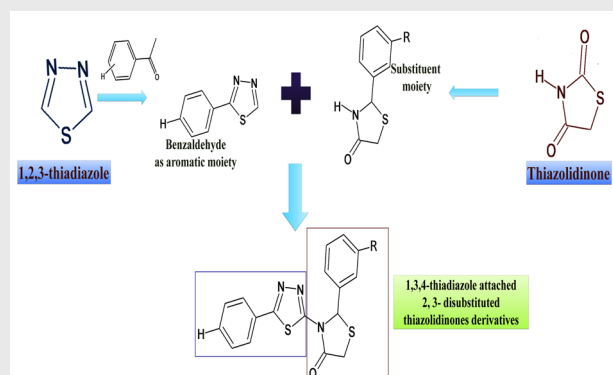
The authors declare no conflict of interest.

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PICTORIAL ABSTRACT



SUMMARY

- A set of conjugates (derivatives) were designed and synthesized and performed for anti-inflammatory, antimicrobial activity and anticancer activity.
- Synthesized derivative shows low toxicity and suggest potential agents for anti-inflammatory and antimicrobial activity.
- The data also suggested the effective anticancer potential of the synthesized derivatives, molecular docking and ADMET prediction studies supports the biological data.

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